

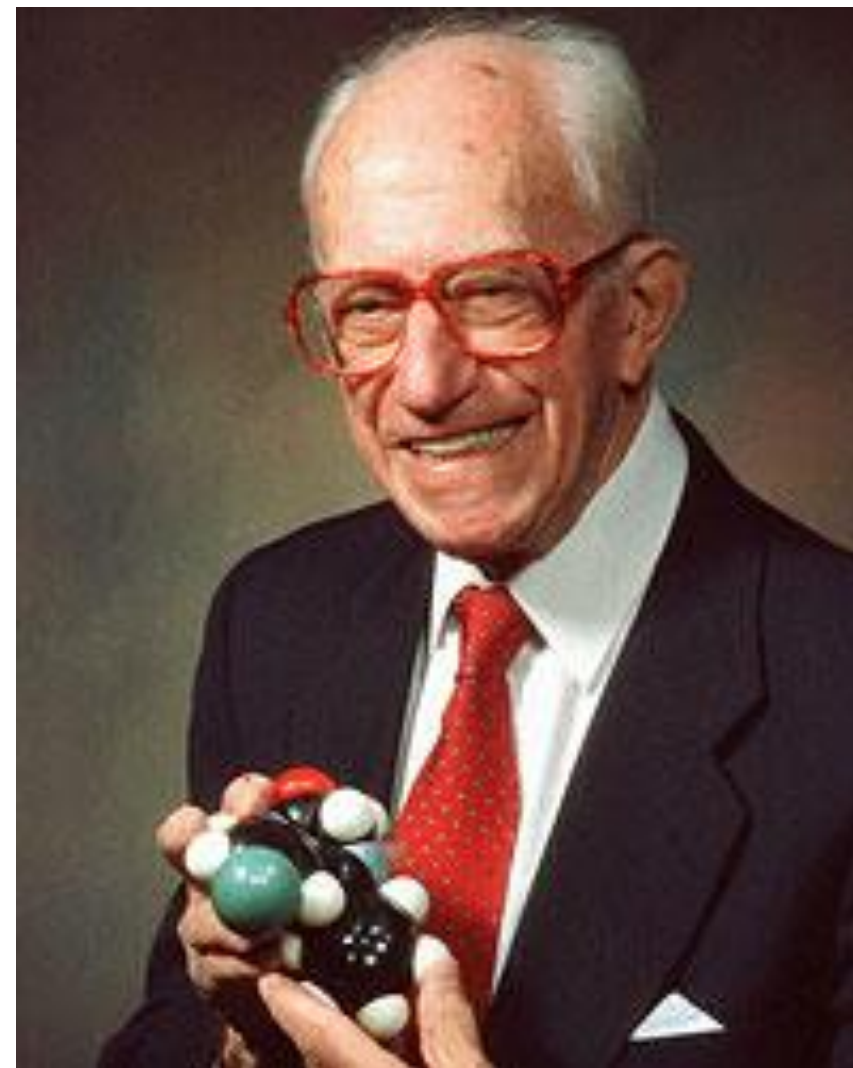
STRUCTURE and ACTIVITY of DRUGS - practical aspects II.

György Domány

Scientific adviser
Gedeon Richter Plc.

Leo H. Sternbach
(1908-2005)

The Benzodiazepine Story
J. Med. Chem. 22, 1-7 (1979)



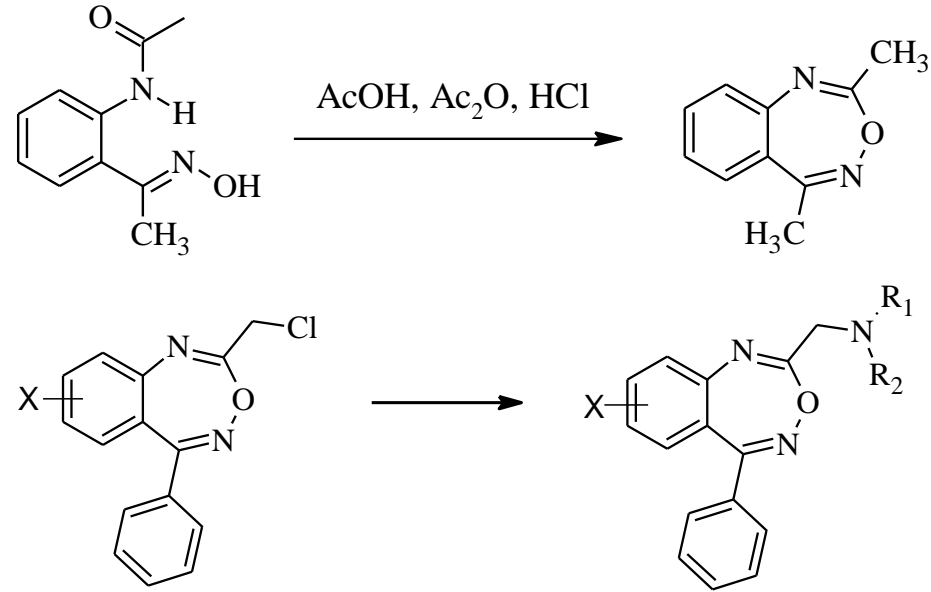
„The pharmacological tests for the screening of sedatives and tranquilizers were well in hand, and we chemists were asked to produce a new compound which would be superior to the then existing tranquilizers.“

„Since our main interest was chemical synthesis we planned to select an approach which would be chemically most attractive, challenging, and satisfying. This left us essentially with two alternatives: to modify existing drugs or to search for a new class of tranquilizers.“

„The class of compounds we were seeking would be expected to fulfill the following criteria:

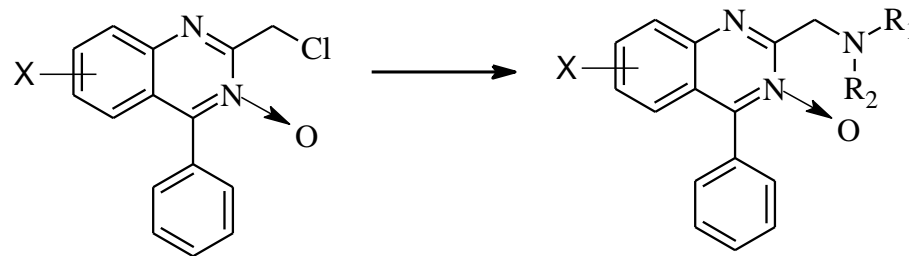
- (1) be relatively unexplored,
- (2) be readily accessible,
- (3) give the possibility of a multitude of variations and transformations,
- (4) offer some challenging chemical problems, and
- (5) „look“ as if it could lead to biologically active products.“

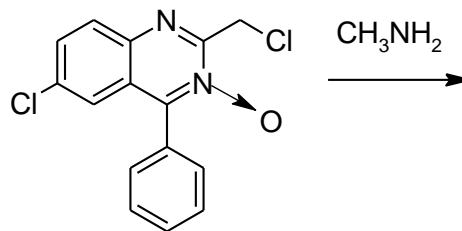
Benzheptodiazines



„The reaction products, we hoped, might have interesting properties, since it is known that basic groups frequently impart biological activity.“

Quinazoline 3-oxides

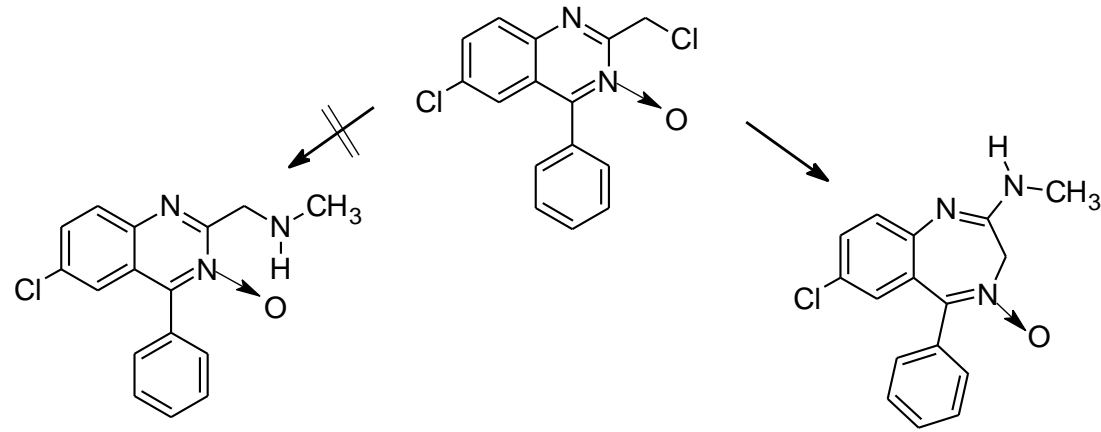




compound	Inclined screen	Foot shock	Cat	Pentylene tetrazole	Anticonvulsant tests	
					electroshock	
					max	min
New comp.	100	40	2	18	92	150
meprobamate	250	250	100	150	200	167
chlorpromazine	17	20	2.5	42	150	600
phenobarbital	120	80	10	75	18	90

Dose (mg/kg) of orally administered drug required to achieve the desired effect.

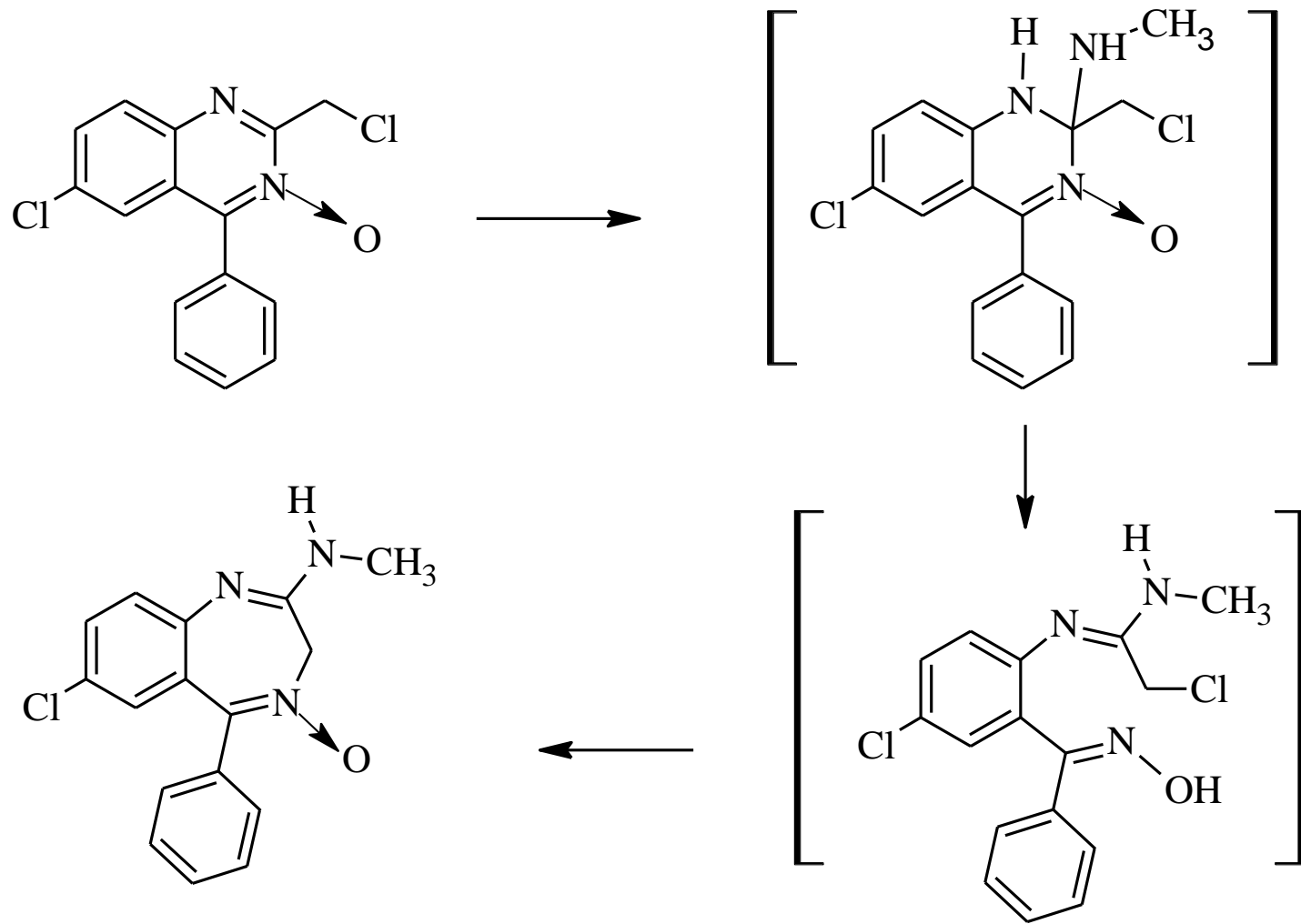
Benzodiazepines



compound	Inclined screen	Foot shock	Cat	Pentylene tetrazole	Anticonvulsant tests	
					electroshock	
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New comp.	100	40	2	18	92	150
meprobamate	250	250	100	150	200	167
chlorpromazine	17	20	2.5	42	150	600
phenobarbital	120	80	10	75	18	90

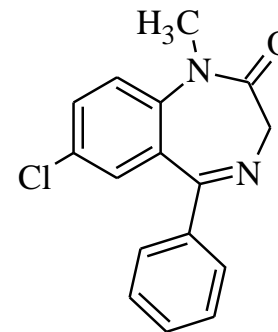
Dose (mg/kg) of orally administered drug required to achieve the desired effect.

Prep.: 1955; test.: May 1957; patent application: May 1958; patent granted: July 1959; launch: 1960 (Librium - Elenium)

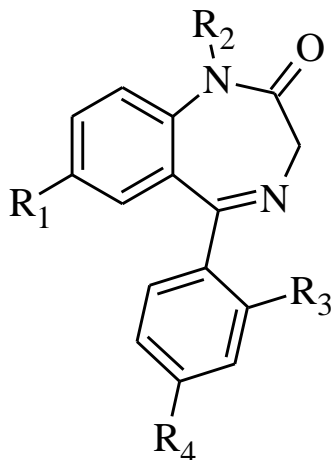


Benzodiazepines

Prepared and tested: 1959; launch: 1963 Valium - Seduxen,
„Mother's Little Helper“ (1966 Rolling Stones)



compound	Inclined screen	Foot shock	cat	Pentylene tetrazole	Anticonvulsant tests	
					electroshock	
					max	min
Librium	100	40	2	18	92	150
Valium	30	10	0.2	1.4	6.4	64



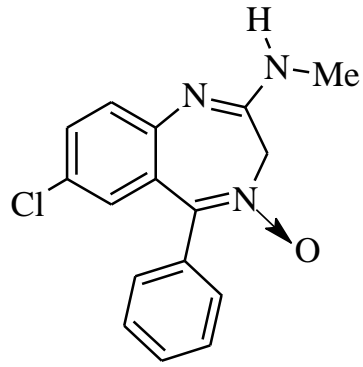
Over 3000 1,4-benzo- and heterodiazepinones by Roche

R_1 : activity increased by electronwithdrawing groups (Hlg, NO_2 , CF_3);

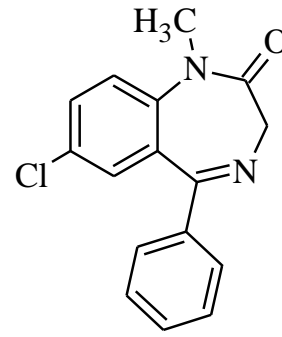
R_2 : activity increased by Me (decreased by larger groups);

R_3 : activity increased by halogens (F, Cl);

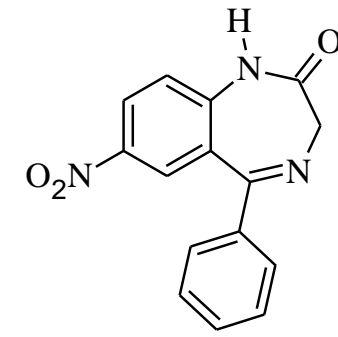
R_4 : activity very strongly decreased by any substituents



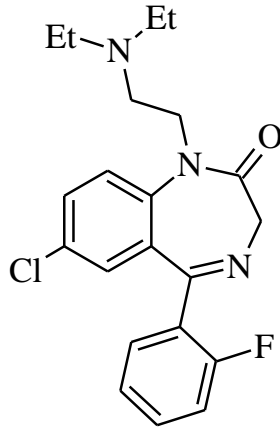
1960 chlordiazepoxide



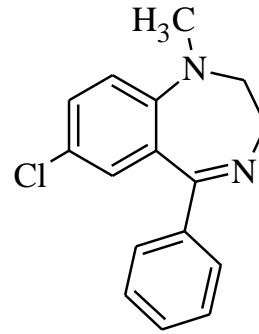
1963 diazepam



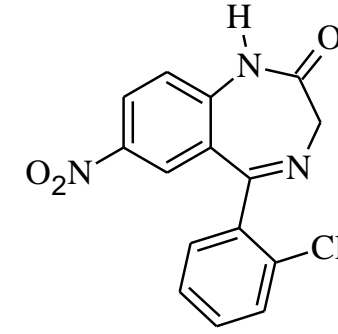
1965 nitrazepam



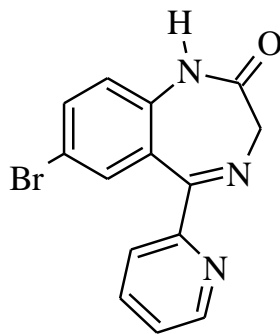
1968 flurazepam



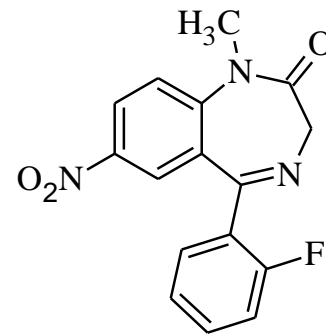
1968 medazepam



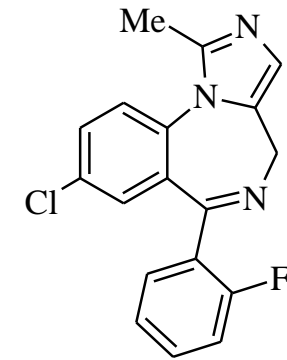
1973 clonazepam



1974 bromazepam



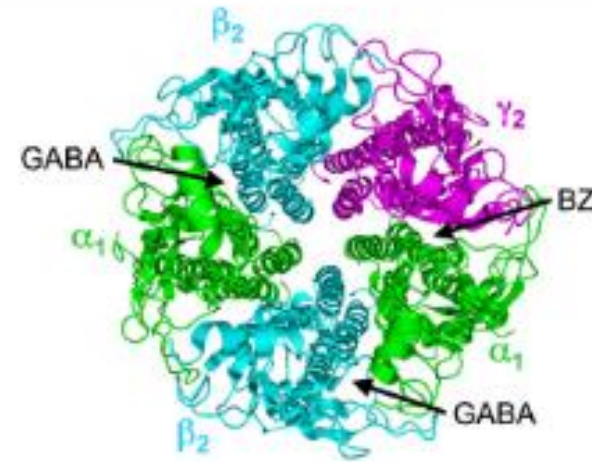
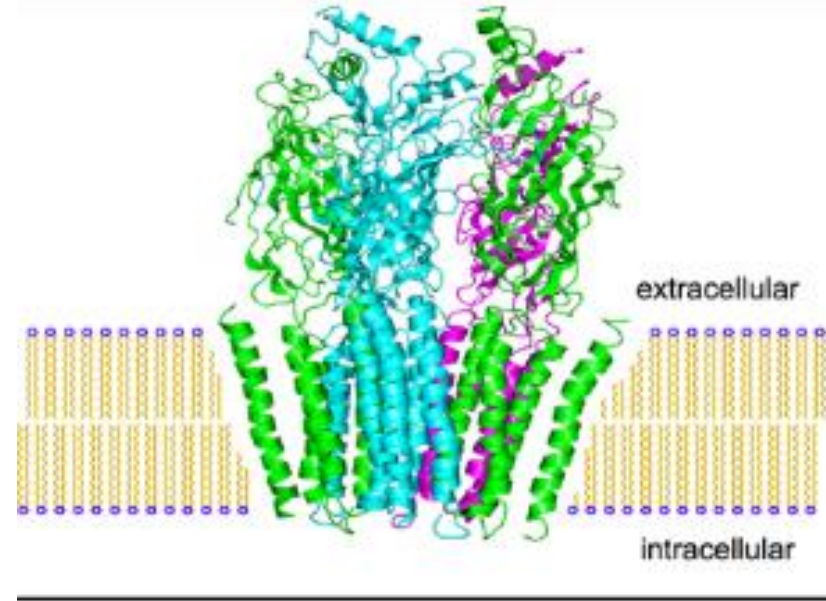
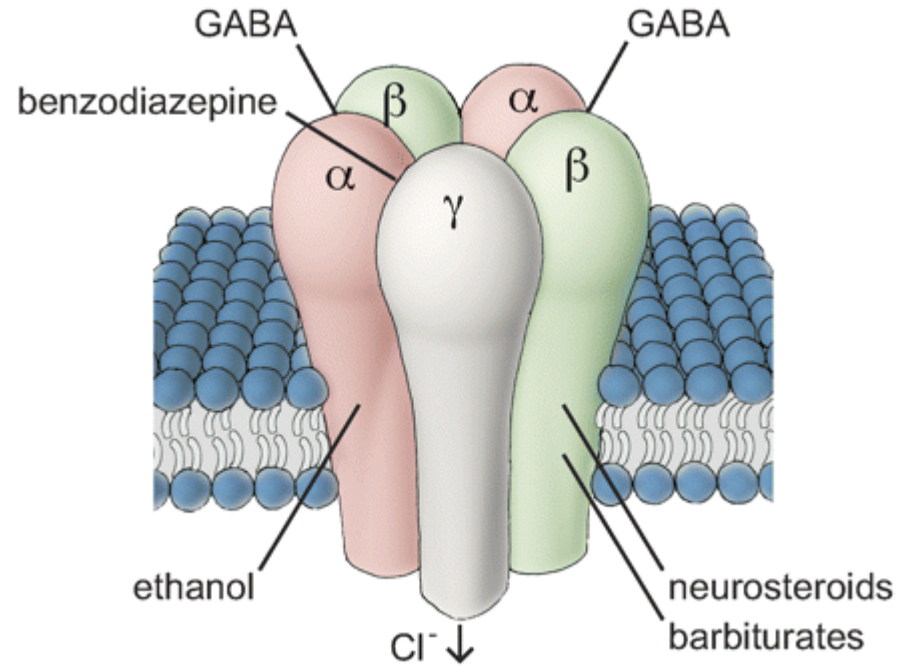
1974 flunitrazepam



1982 midazolam

1960 chlordiazepoxide (Roche)	1963 diazepam (Roche)	1964 oxazepam (Pfizer)
1965 nitrazepam (Roche)	1967 clorazapate (Sanofi)	1968 flurazepam (Roche)
1968 medazepam (Roche)	1969 temazepam (Pfizer)	1971 lorazepam (Pfizer)
1973 clonazepam (Roche)	1973 nordazepam (Midy)	1973 prazepam (Pfizer)
1974 bromazepam (Roche)	1974 flunitrazepam (Roche)	1974 tofizopam (Mochida)
1975 clomazam (Sanofi)	1975 estazolam (Abbott)	1978 triazolam (Pfizer)
1979 clotiazepam (Mitsubishi)	1980 ketazolam (Pfizer)	1980 lormetazepam (Pfizer)
1982 ethyl loflazepate (Sanofi)	1982 halazepam (Merck&Co.)	1982 midazolam (Roche)
1983 alprazolam (Pfizer)	1983 loprazolam (Sanofi)	1984 brotizolam (Boehringer)
1984 doxefazepam (Schiapp.)	1985 quazepam (Merck&Co.)	1992 cinolazepam (GL Pharma.)

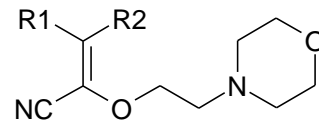
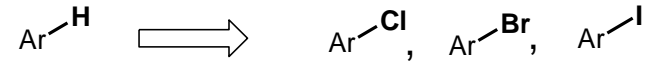
MoA: $GABA_A$ positive allosteric modulators



Bioisosterism: the relationship between bioisosteres, substituents or groups with similar physical or chemical properties that impart similar biological properties to a chemical compound.

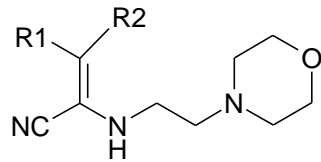
Bioisosterism

A

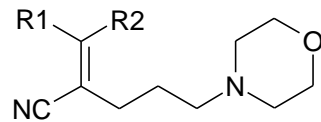
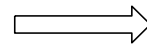


PDE4 IC₅₀: 3,8 nM

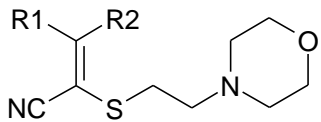
B



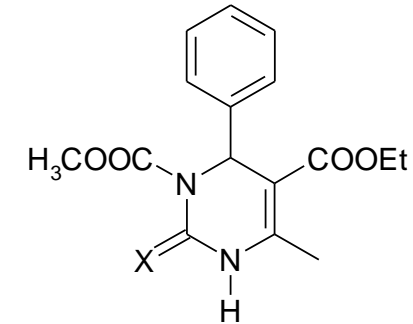
PDE4 IC₅₀: 8,2 nM



PDE4 IC₅₀: 1,4 nM



PDE4 IC₅₀: 0,6 nM

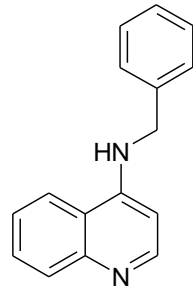


X	Van der Waal's radiusz (Å)	IC ₅₀ (nM)*
O	1,40	140
NH	1,50	160
S	1,85	17

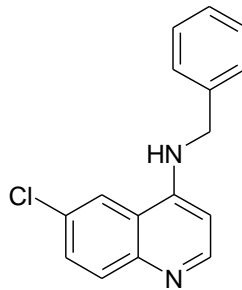
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Bioisosterism

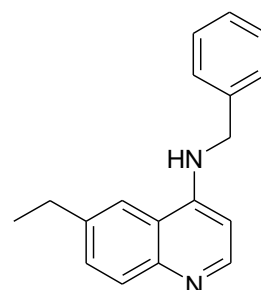
Adenosin A3 antagonists



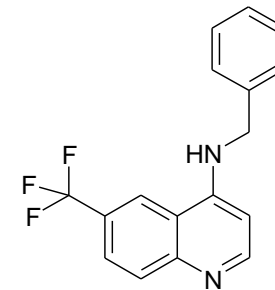
clogP: 3,41 (szám.)



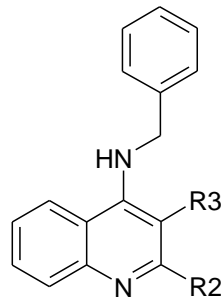
clogP: 4,07 (szám.)



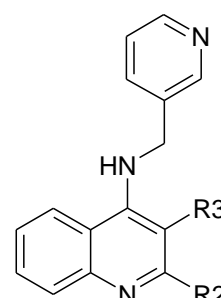
clogP: 4,35 (szám.)



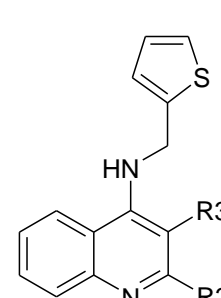
clogP: 4,35 (szám.)



A3 IC₅₀: 0,5 nM



A3 IC₅₀: 1,7 nM

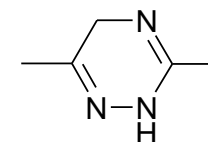
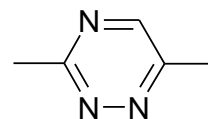
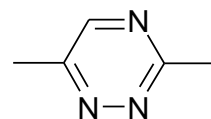
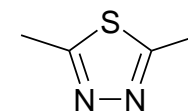
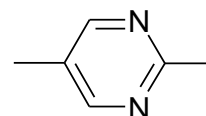
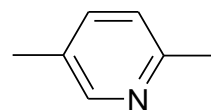
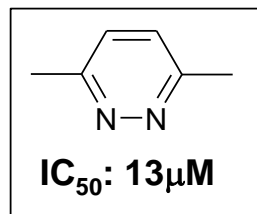
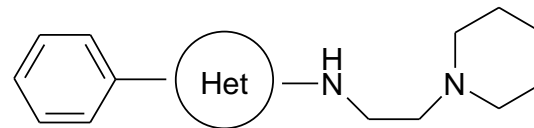


A3 IC₅₀: 0,4 nM

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Bioisosterism

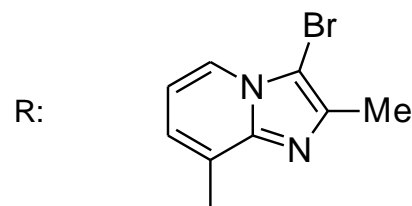
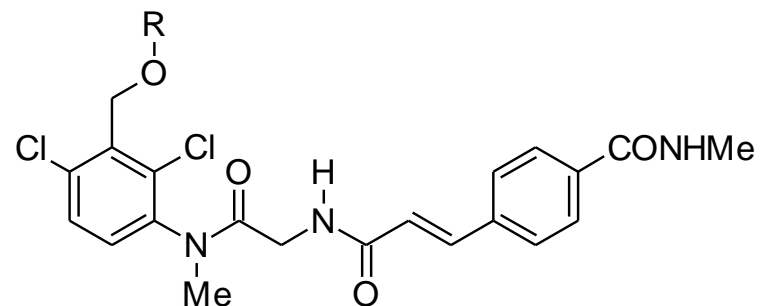
AChE inhibitors



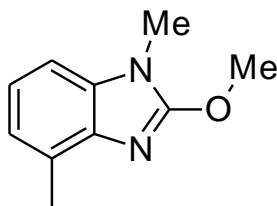
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Bioisosterism

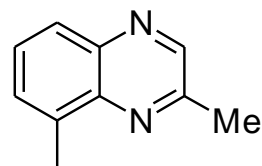
Bradykinin B2 antagonists



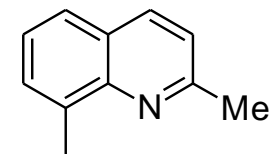
IC₅₀: 1,5 nM



IC₅₀: 2,2 nM



IC₅₀: 3,1 nM

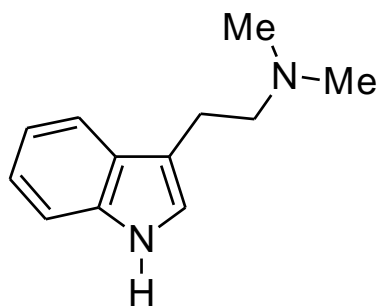


IC₅₀: 1,3 nM

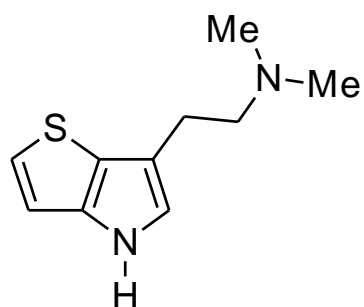
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Bioisosterism

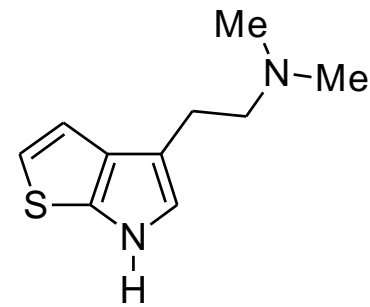
5HT receptor ligands



A



B



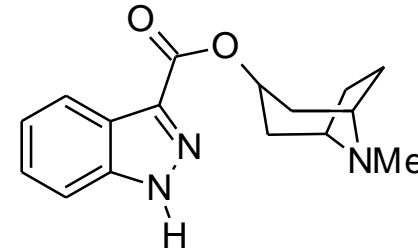
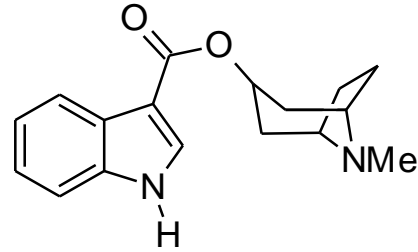
C

	5HT _{1A} K _i (nM)	5HT _{2A} K _i (nM)	5HT _{2B} K _i (nM)	5HT _{2C} K _i (nM)
A	259	65	101	33
B	76	276	214	64
C	184	106	483	102

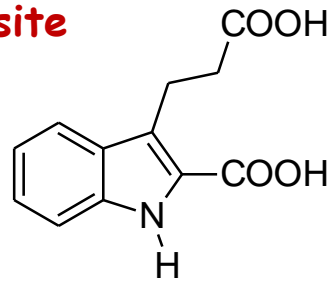
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Bioisosterism

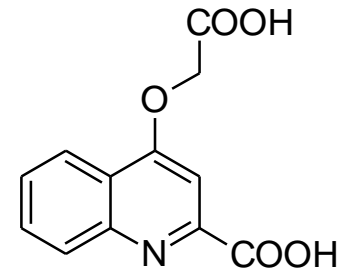
5HT₃ agonists



NMDA glicin site



Gly IC₅₀: 27 μM

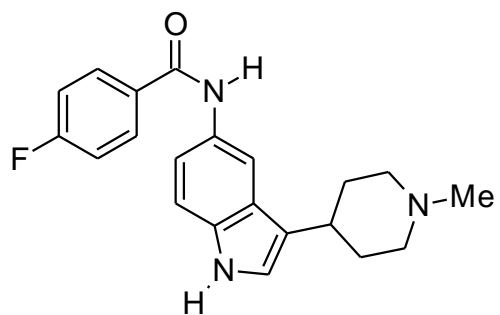


Gly IC₅₀: 25 μM

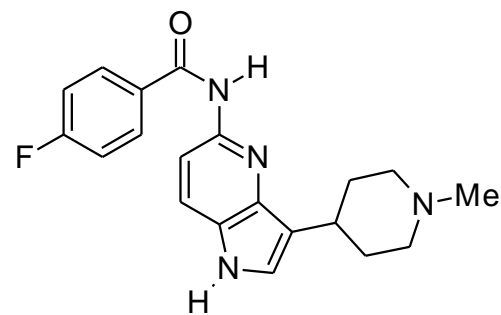
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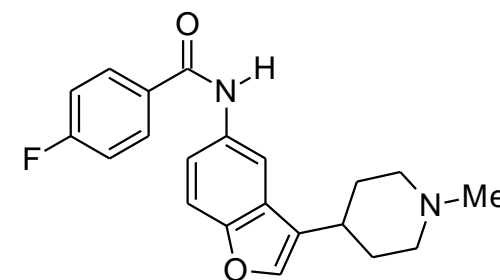
5HT_{1F} receptor agonists



1



2



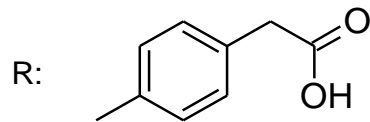
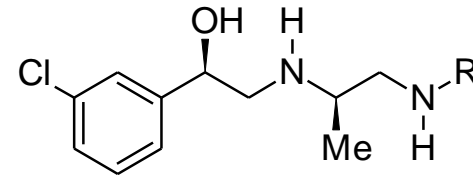
3

	5HT _{1F} Ki (nM)	5HT szelektivitás		
		1A/1F	1B/1F	1D/1F
1	1,6	7	85	86
2	7,6	7,3	160	960
3	3,1	134	>1000	>1000

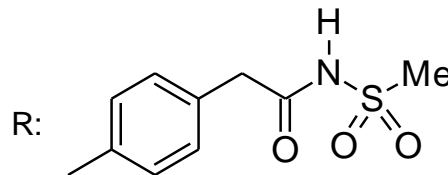
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Bioisosterism

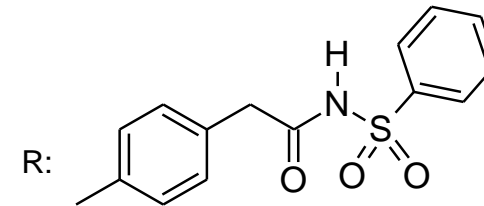
β 3 adrenoceptor agonists



pEC_{50} : 7,8
 E_{max} : 117%



pEC_{50} : 8,3
 E_{max} : 96%

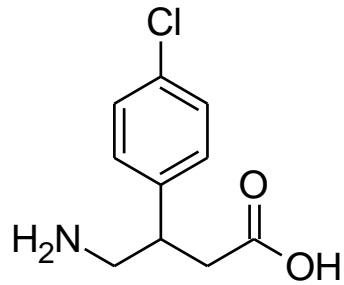


pEC_{50} : 9,1
 E_{max} : 110%

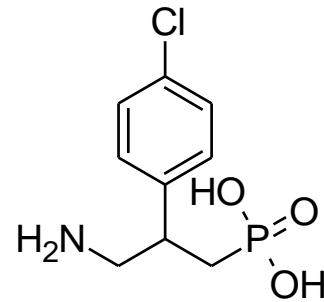
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Bioisosterism

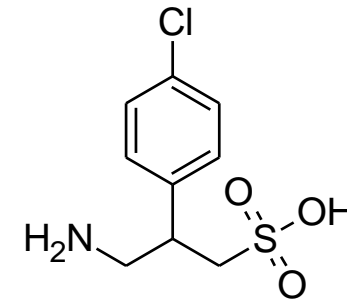
GABA_B agonists



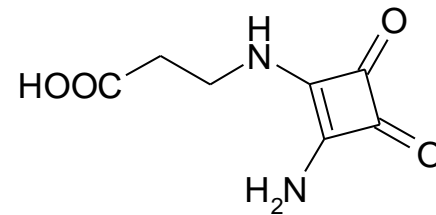
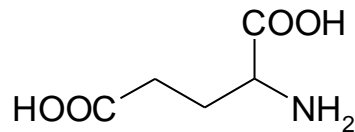
baclofen



phaclofen



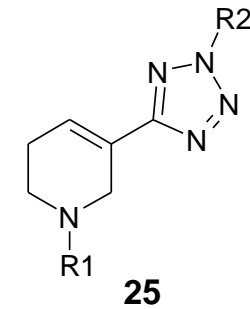
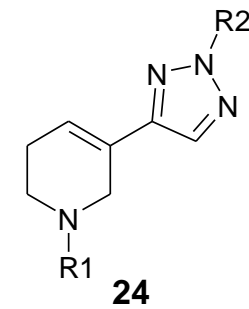
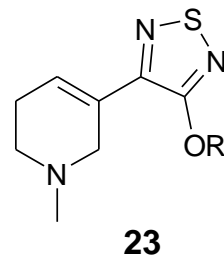
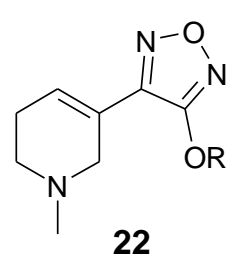
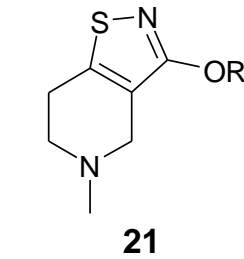
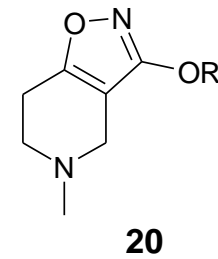
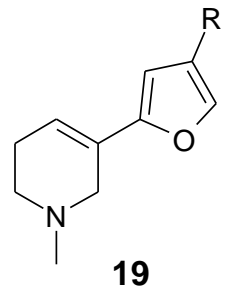
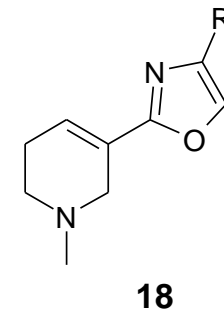
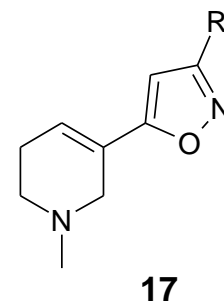
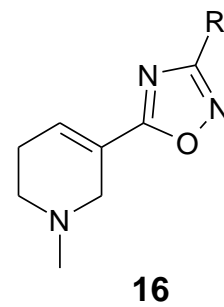
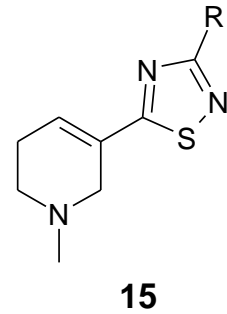
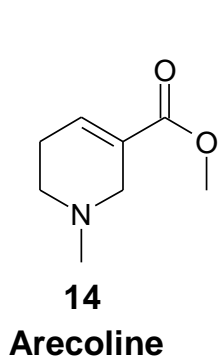
saclofen



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Bioisosterism

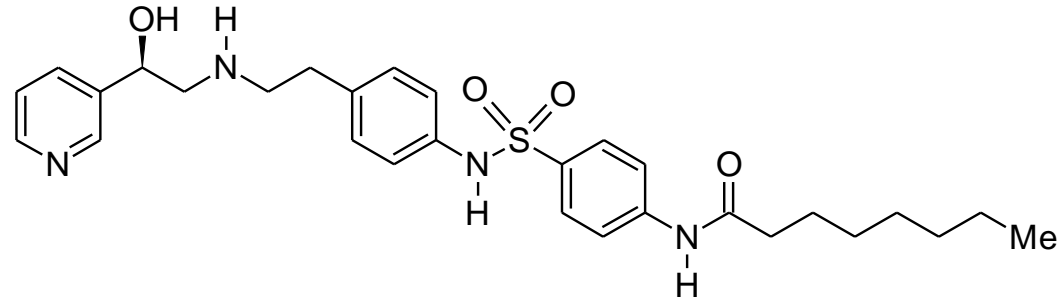
Muscarinic agonists



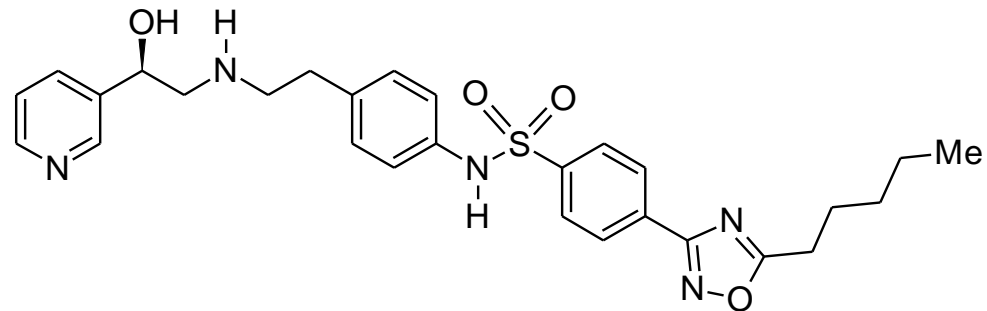
Keserű György Miklós *A gyógyszerkutató kémia* Akadémiai Kiadó, Budapest, 2011

Bioisosterism

β_3 adrenoceptor agonists



β_3 EC_{50} : 18 nM, E_{max} : 81%

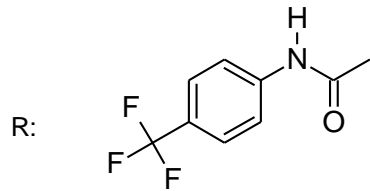
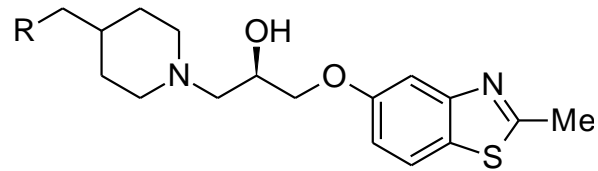


β_3 EC_{50} : 23 nM, E_{max} : 53%

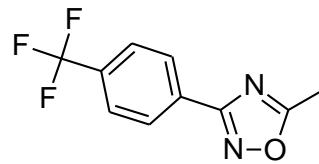
Keserű György Miklós *A gyógyszerkutató kémia* Akadémiai Kiadó, Budapest, 2011

Bioisosterism

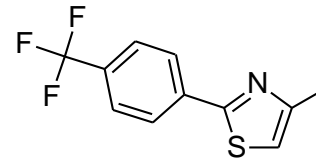
Fatty acid oxidation inhibitors



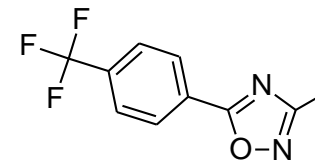
IC₅₀: 7,6 μM
met.stab.: 28%



IC₅₀: 3,6 μM
met.stab.: 72%



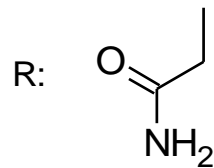
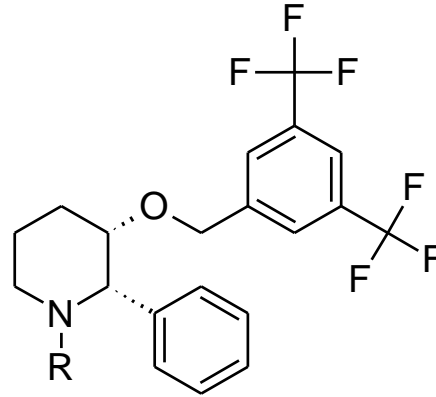
IC₅₀: 1,3 μM
met.stab.: 40%



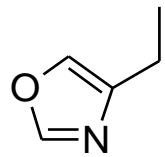
IC₅₀: 0,38 μM
met.stab.: 74%

Bioisosterism

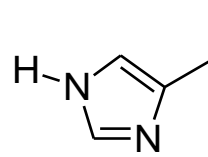
NK1 antagonists



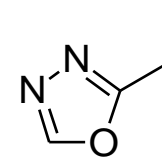
IC_{50} : 1,3 nM



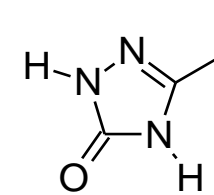
IC_{50} : 1,0 nM



IC_{50} : 0,90 nM



IC_{50} : 0,97 nM

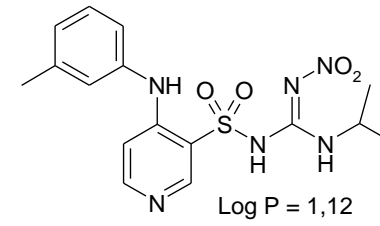
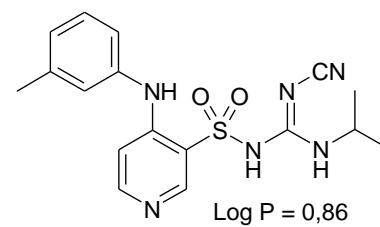
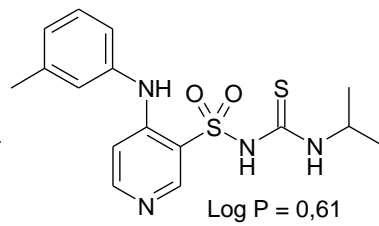
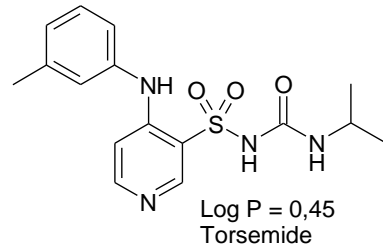


IC_{50} : 0,05 nM

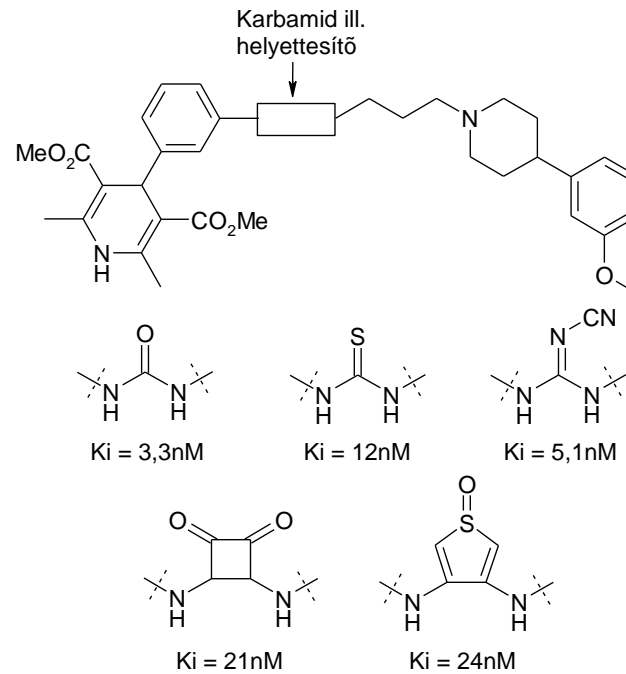
Keserű György Miklós *A gyógyszerkutató kémia* Akadémiai Kiadó, Budapest, 2011

Bioisosterism

Diuretics



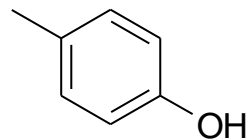
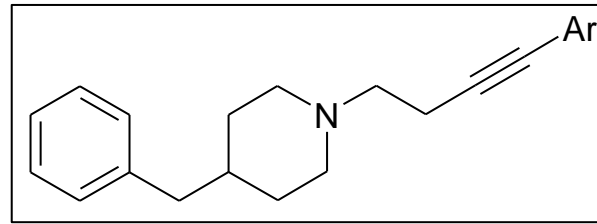
NPY Y1 antagonists



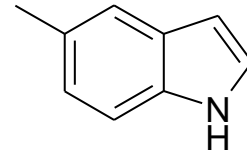
Keserű György Miklós *A gyógyszerkutató kémia* Akadémiai Kiadó, Budapest, 2011

Bioisosterism

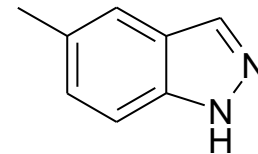
NMDA antagonists



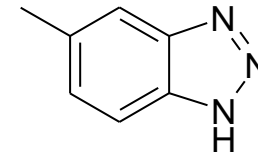
4-Fenol
 IC_{50} : 0,17 μ M



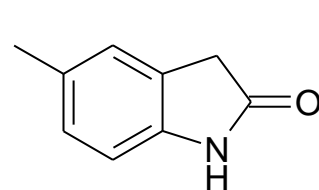
5-Indol
 IC_{50} : 0,63 μ M



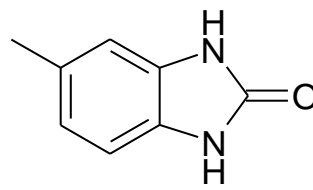
5-Indazol
 IC_{50} : 0,25 μ M



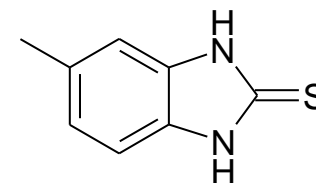
5-Benzotriazol
 IC_{50} : 0,22 μ M



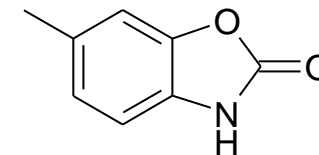
5-Indolon
 IC_{50} : 0,32 μ M



5-Imidazolone
 IC_{50} : 0,09 μ M



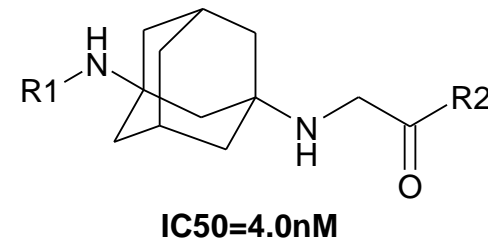
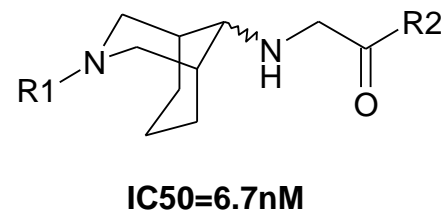
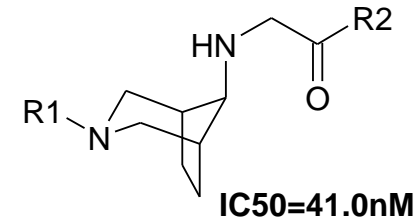
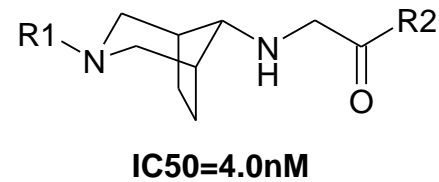
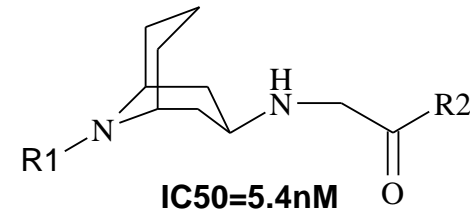
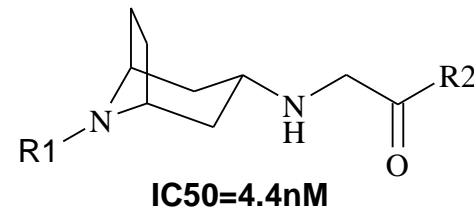
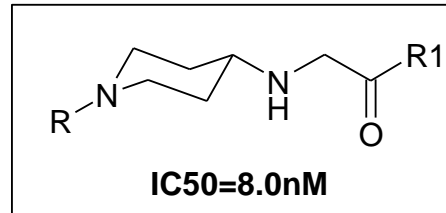
5-Imidazolinone
 IC_{50} : 0,18 μ M



5-Imidazolone
 IC_{50} : 0,12 μ M

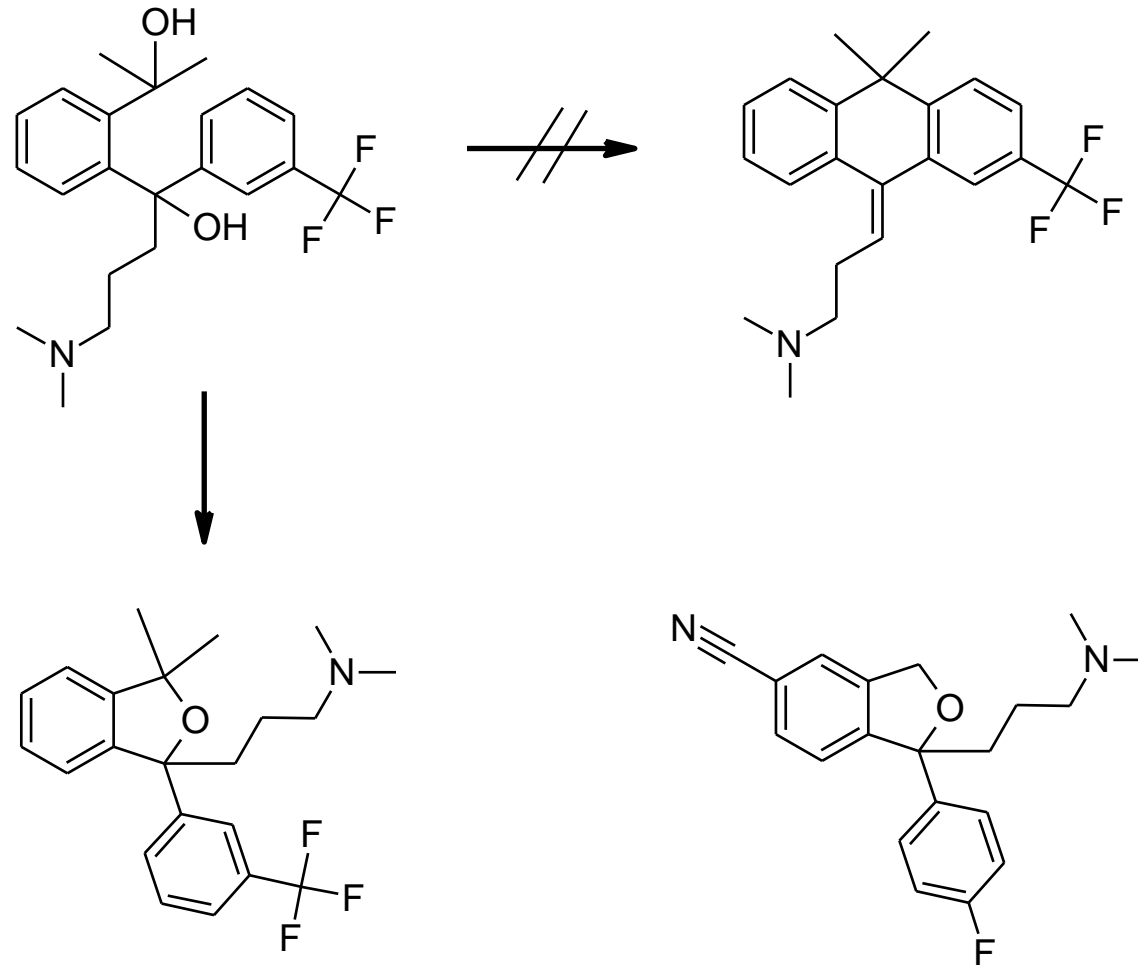
Bioisosterism

DPP IV inhibitors



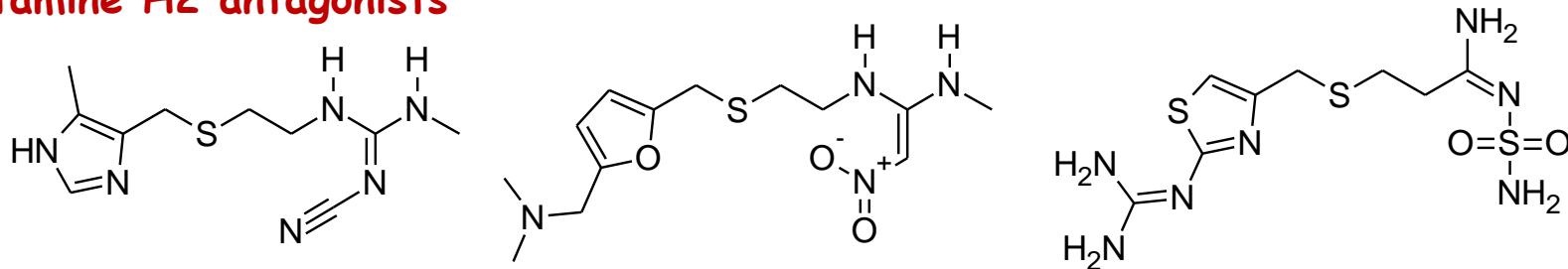
Keserű György Miklós *A gyógyszerkutató kémia* Akadémiai Kiadó, Budapest, 2011

Scaffold hopping

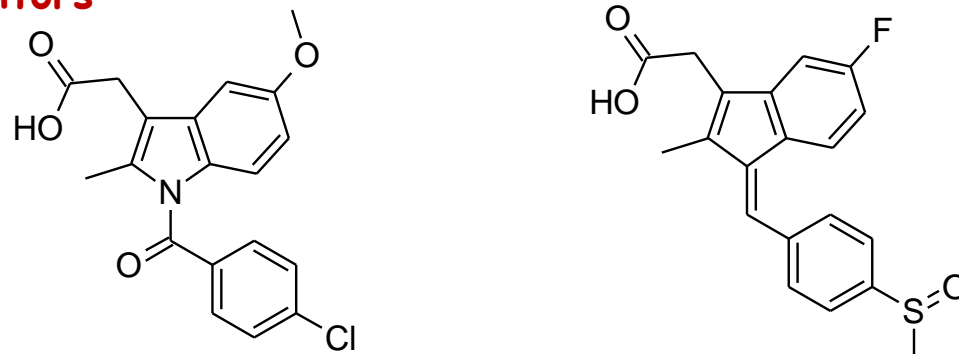


Scaffold hopping

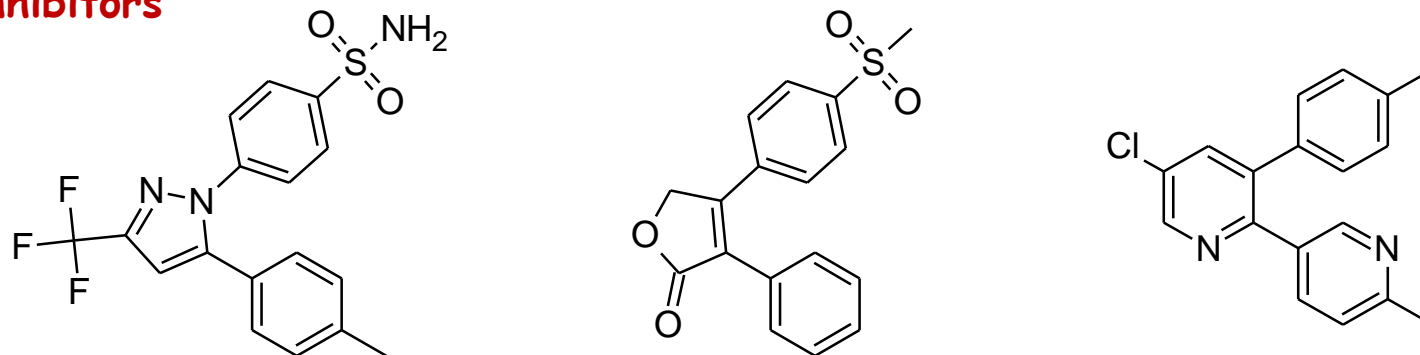
Histamine H2 antagonists



COX1/COX2 inhibitors

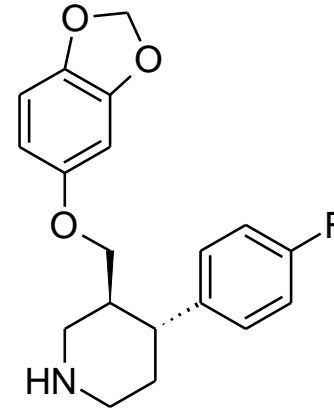
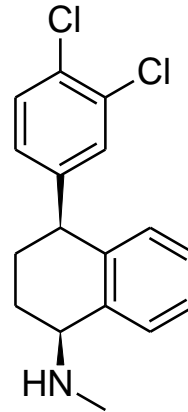
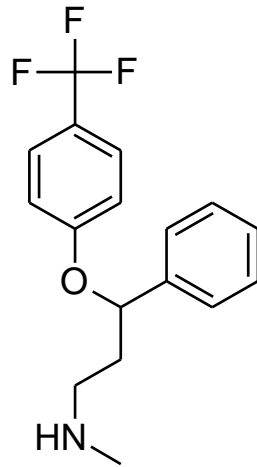


COX2 inhibitors

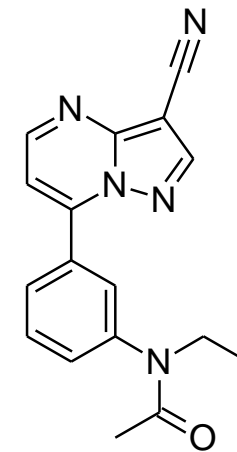
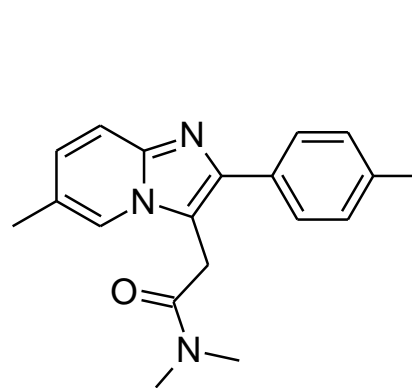
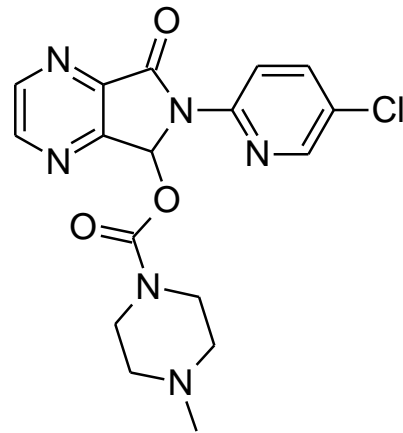


Scaffold hopping

SSRIs

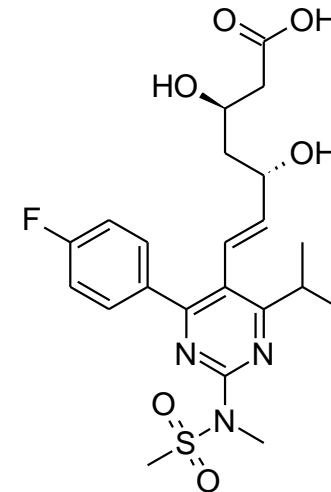
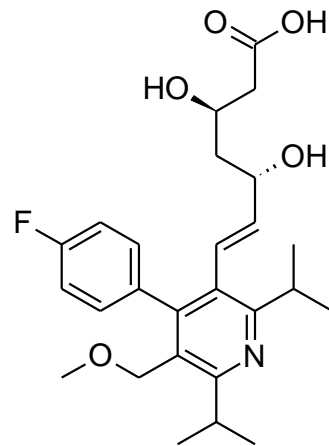
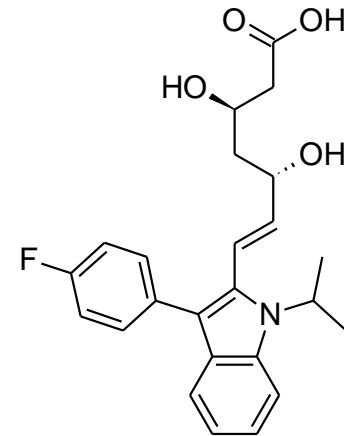
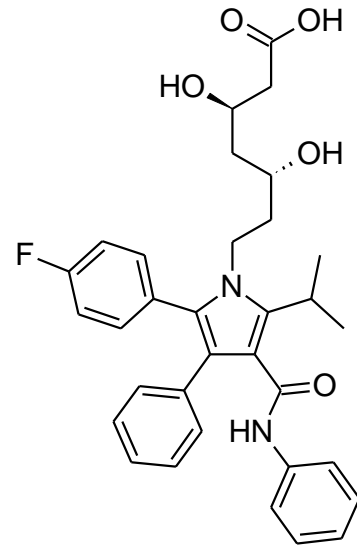


GABA_A ligands



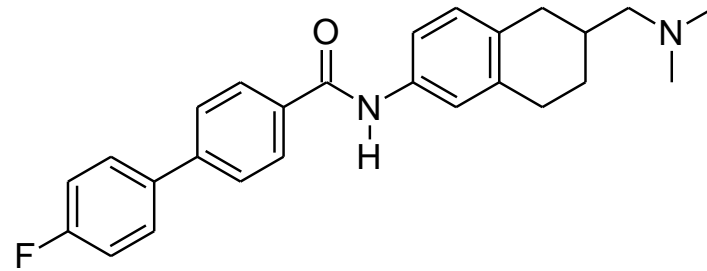
Scaffold hopping

HMG-CoA reductase inhibitors

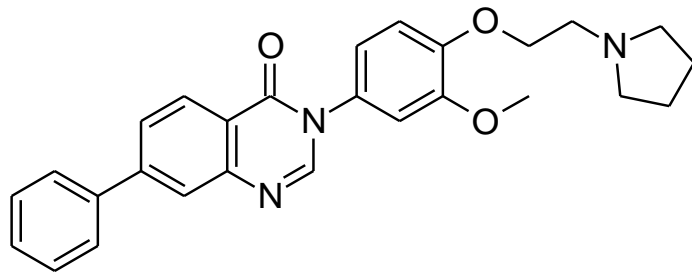


Scaffold hopping

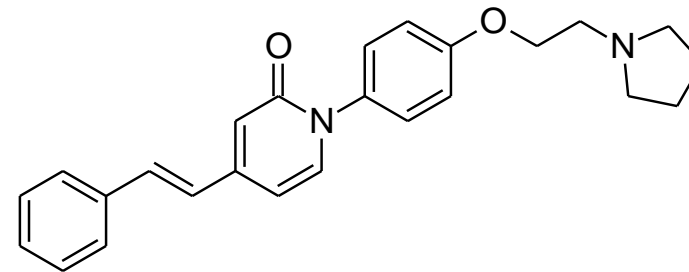
MCHR-1 antagonists



GTPγS IC₅₀: 19 nM



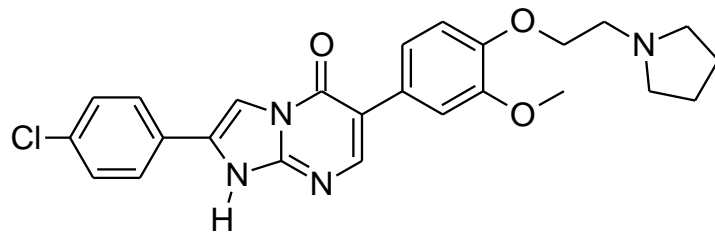
K_i: 2,1 nM (Boehringer)



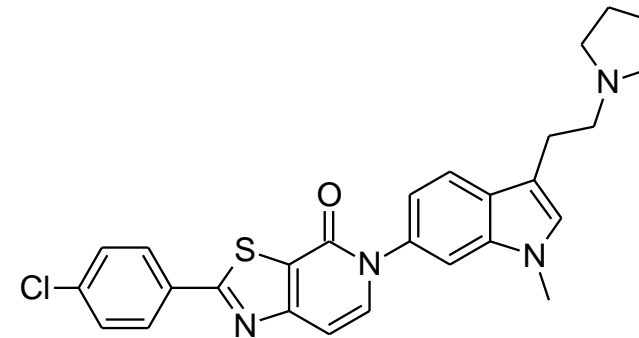
K_i: 1,5 nM (Banyu)

Scaffold hopping

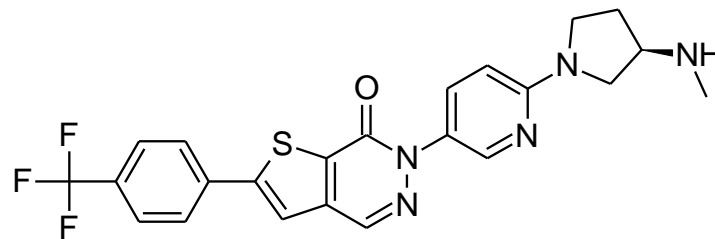
MCHR-1 antagonists



K_i : 77 nM (Pharmacopeia)



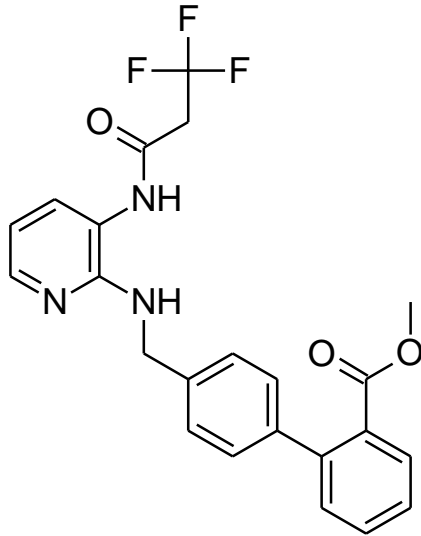
K_i : 3,16 nM (Eli Lilly)



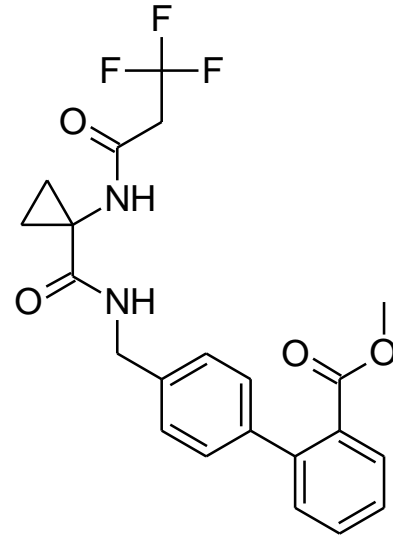
K_i : 3,3 nM (Neurocrine)

Scaffold hopping

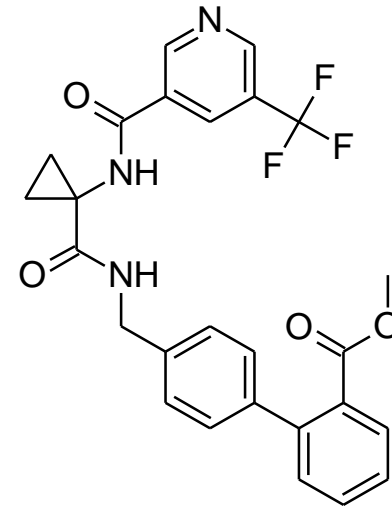
Bradykinin B1 antagonists



K_i : 11.8 nM



K_i : 63 nM

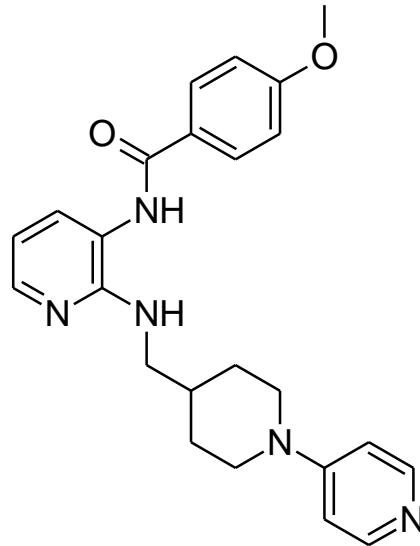


K_i : 1.8 nM

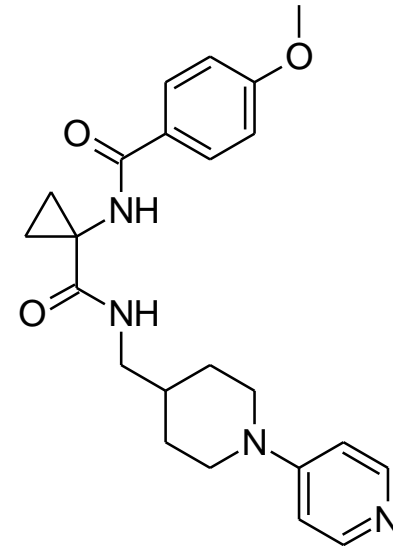
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Scaffold hopping

Factor Xa inhibitors



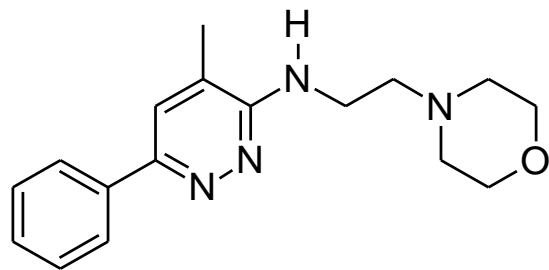
K_i : 39 nM



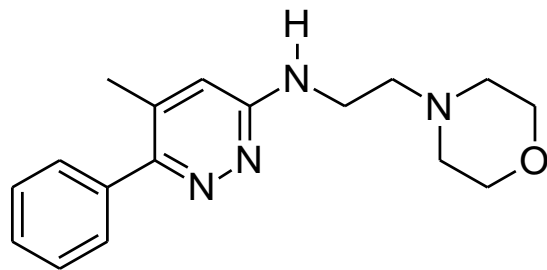
K_i : 175 nM

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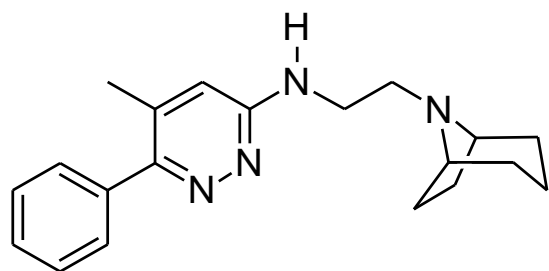
Selective Optimization of Side Activities (SOSA)



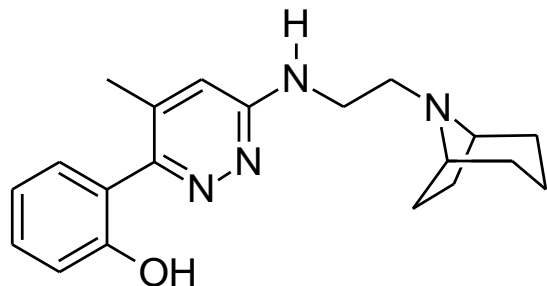
$M_1 K_i$: 17000 nM
Minaprine (MAO-A)



$M_1 K_i$: 550 nM



$M_1 K_i$: 50 nM



$M_1 K_i$: 3 nM



Camille G. Wermuth (1933-2015)

Scaffold hopping: practical application of bioisosterism

Druglikeness, structure/property relationships

Christopher A. Lipinski

Born: 1944 (?)

BSc: 1965

PhD: 1968

Pfizer: 1970-2002

„Rule of 5“: 1995



Goal: to identify compounds that reached Phase II clinical trial

World Drug Index : 50427 compounds

Out of which United States Adopted Name: 7894

Out of which International Non-proprietary Name: 6320

The sum of the above two groups minus the identicals: 8548

„Indication and usage“ suggesting clinical exposure: 3704

Minus 1176 „POLY“, 87 „PEPTIDE“, 101 „Quat“, 53 O=P-O.

REMAINED 2245-MEMBERED COMPOUND LIBRARY CALLED „USAN“

Calculated properties influencing solubility/permeability

Molecular weight: 11% of „USAN“ had >500;

Lipophilicity (octanol/water partition coefficient, LogP): 10% of „USAN“ had > 5;

H-bond donors (OH+NH): 8% of „USAN“ had > 5;

H-bond acceptors (O+N): 12% of „USAN“ had > 10.

„The 'rule of 5' states that: poor absorption or permeation are more likely when:

There are more than 5 H-bond donors (expressed as the sum of OHs and NHs);

The MWT is over 500;

The Log P is over 5 (or MLogP is over 4.15);

There are more than 10 H-bond acceptors (expressed as the sum of Ns and Os)

Compound classes that are substrates for biological transporters are exceptions to the rule.”

The ,Rule of 5' was implemented at Pfizer in 1995, and published in 1997.

Advanced Drug Delivery Reviews Volume 23, Issue 1-3, 15

January 1997, Pages 3-25

Experimental and computational approaches to estimate solubility and permeability in drug discovery and development settings (Review)

Lipinski, C.A. Email Author,

Lombardo, F.,

Dominy, B.W.,

Feeney, P.J.

Cited by: 5377

This paper was republished in a special issue dedicated to Dr. Eric Tomlinson, Advanced Drug Delivery Reviews, A Selection of the Most Highly Cited Articles, 1991-1998

Advanced Drug Delivery Reviews Volume 46, Issue 1-3, 1 March 2001, Pages 3-26

Experimental and computational approaches to estimate solubility and permeability in drug discovery and development settings (Article)

Lipinski, C.A. Email Author,

Lombardo, F.,

Dominy, B.W.,

Feeney, P.J.

Cited by: 4942

Christopher A. Lipinski

**,Rule of 5' paper
citations: 10319**

Total papers: 60

Total citations: 15974

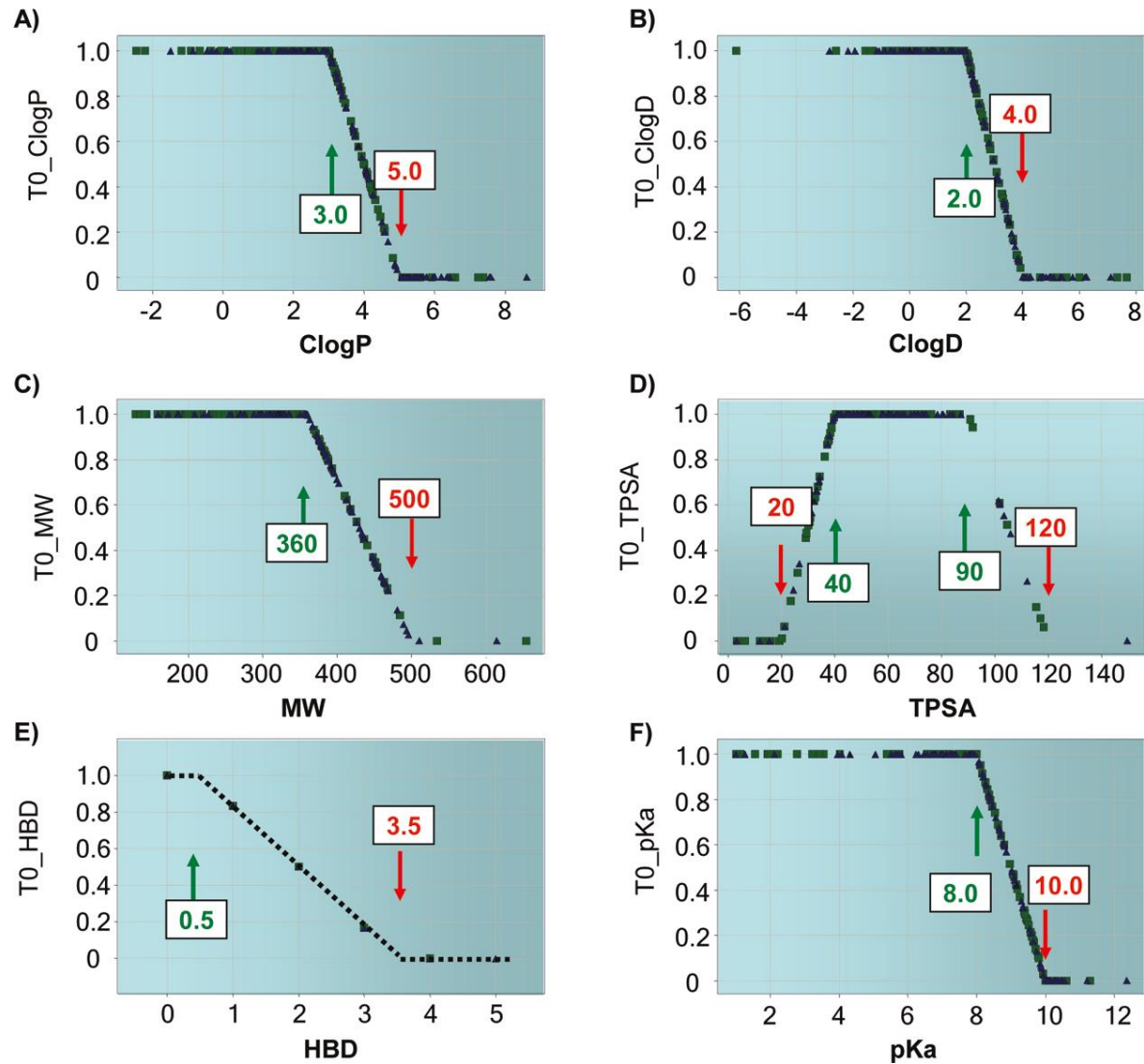


Rules for Rapid Property Profiling from Structure

	MW	logP	HBD	HBA	TPSA
Rule of 5 drugs	< 500	< 5	< 5	< 10	< 140
Rule of 4 leads	< 400	< 4	< 4	< 8	< 120
Rule of 3 fragments	< 300	< 3	< 3	< 3	< 60

Graham F. Smith *Progress in Med. Chem.* **48**, 1-29 (2009)

Central Nervous System Multiparameter Optimization (CNS MPO)¹



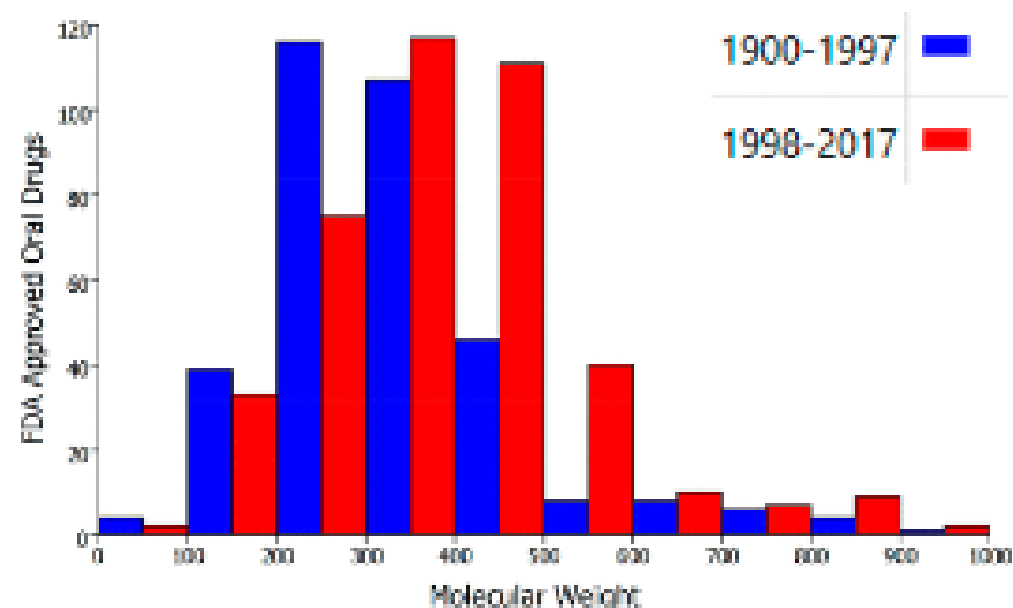
¹Travis T. Wager et al. *ACS Chemical Neuroscience* **1**, 434-449 (2010)

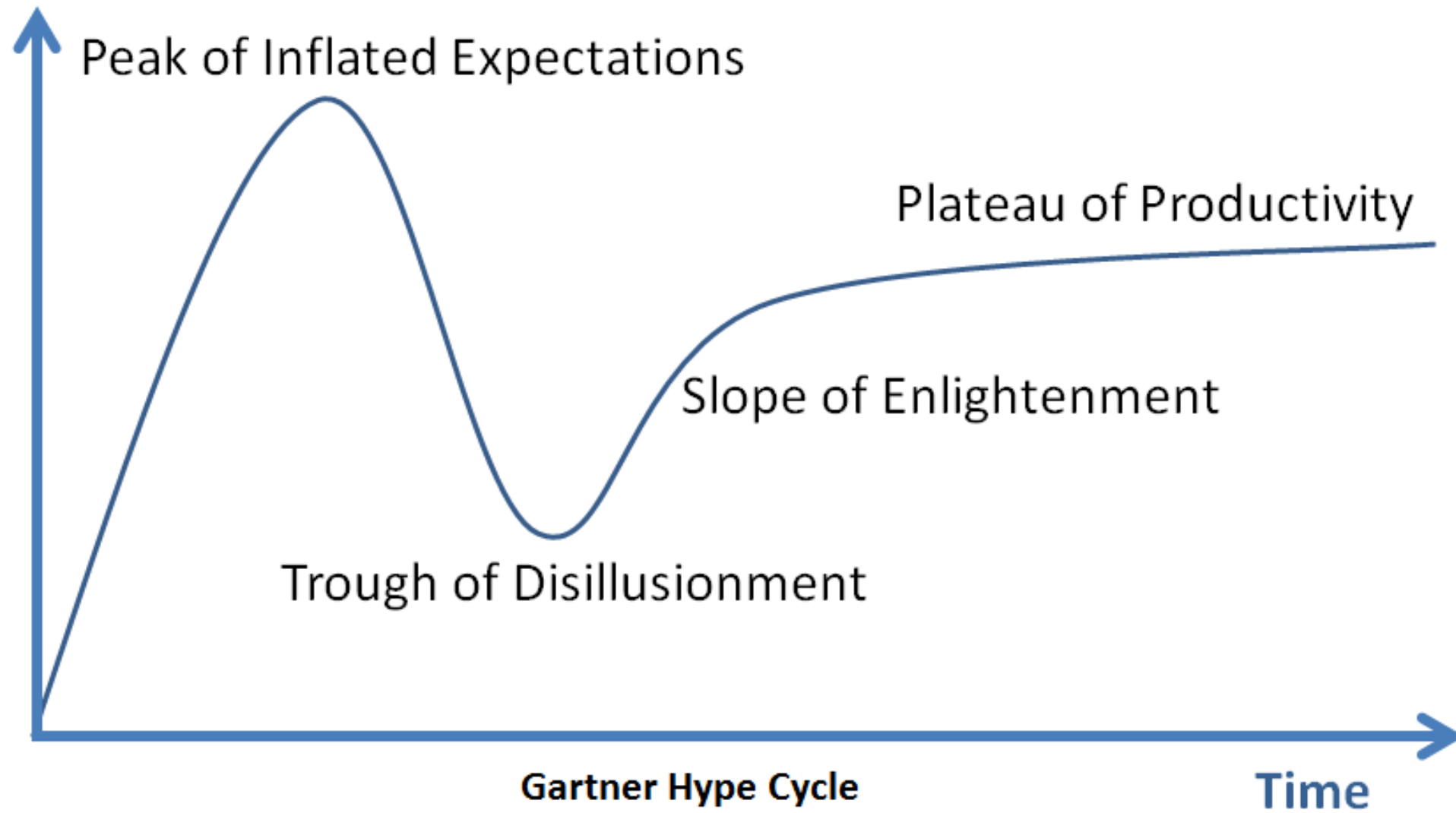
Two Decades under the Influence of the Rule of Five and the Changing Properties of Approved Oral Drugs

Michael D. Shultz

Publication Date (Web): September 13, 2018 (Perspective)

DOI: 10.1021/acs.jmedchem.8b00686





„Rule of 5“,
CNS MPO

Aqueous solubility

Aqueous solubility (s)

„Results indicated that the majority of CNS drugs are highly soluble. Greater than 85% of the drugs have solubility greater than 100 μM ; greater than 90% of the drugs have solubility greater than 10 μM .“

„ when the measured solubility is **less than 1 μM** , the compound is unlikely to become a CNS drug“

„when the measured solubility is **less than 10 μM** , there are high risks associated with the compound advancement“¹

General Solubility Equation by Yalkowsky

$$\log s = 0.5 - \log P - 0.01(\text{m.p.} - 25 \text{ }^\circ\text{C})$$

$$\log s_{\text{pH}7.4} = 0.5 - \log D_{\text{pH}7.4} - 0.01(\text{m.p.} - 25 \text{ }^\circ\text{C})$$

$$[s] = \text{M}; [\text{m.p.}] = \text{ }^\circ\text{C}$$

Bioavailability = f(solubility, permeability)

Solubility range: 0.1 $\mu\text{g/ml}$ - 100 mg/ml ; Permeability range: 0.001 min^{-1} - 0.05 min^{-1}

¹Yun W. Alelyunas et al. *B.M.C.L.* **20**, 7312-7316 (2010)

Structure Modification Strategies for Solubility Improvement

Add ionizable group

Reduce logP

Add hydrogen bonding

Add polar group

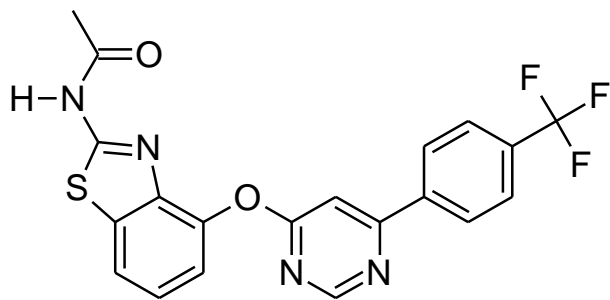
Reduce molecular weight

Out-of-plane substitution to reduce crystal packing

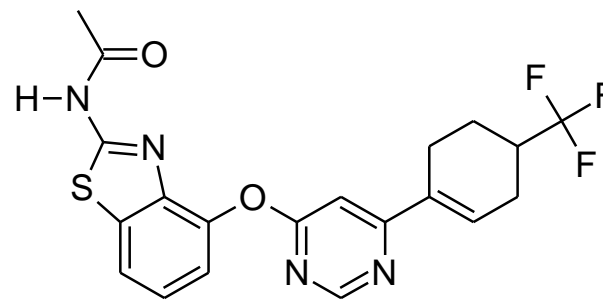
Construct prodrug

Solubility

TRPV1 antagonists



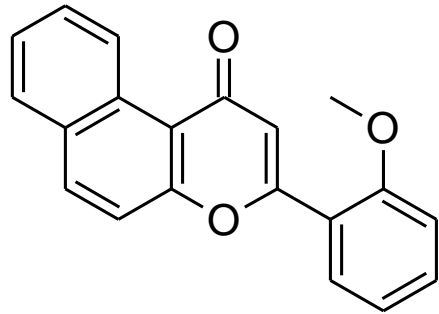
rat TRPV1 (acid) IC_{50} : 0.5 nM
s (0.01 M HCl): < 1 μ g/ml
clogP: 4.6
m.p.: 219-221 °C



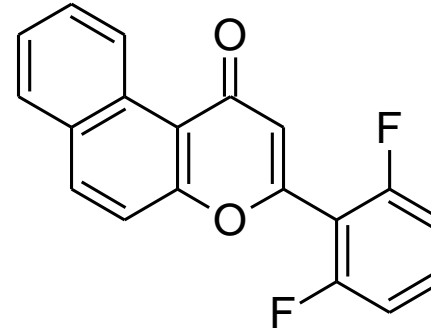
rat TRPV1 (acid) IC_{50} : 2.4 nM
s (0.01 M HCl): 13 μ g/ml
clogP: 3.7
m.p.: 130-131 °C

Solubility

AhR agonists



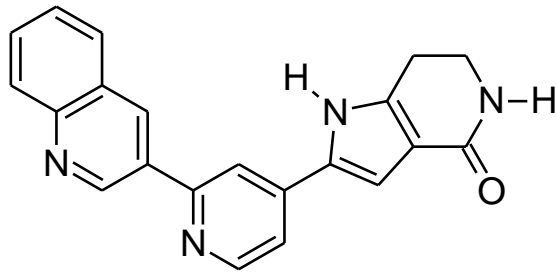
EROD EC₅₀: 0.27 nM
s: 45.8 µg/ml
clogP: 4.1
m.p.: 192-193 °C



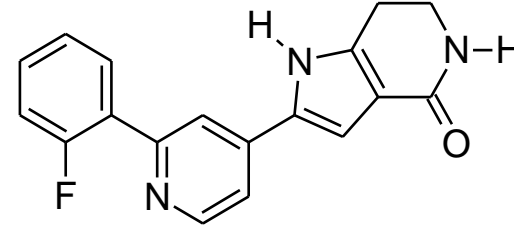
EROD EC₅₀: 0.20 nM
s: 248 µg/ml
clogP: 4.9
m.p.: 150 °C

Solubility

MK-2 inhibitors



MK-2 IC_{50} : 8.5 nM
s: < 0.4 μ M
clogP: 2.7

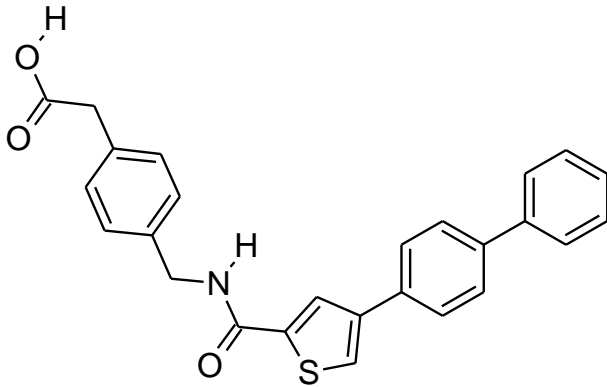


MK-2 IC_{50} : 126 nM
s: 160 μ M
clogP: 2.9

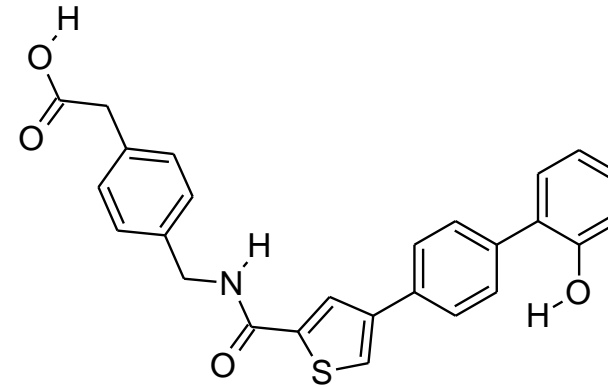
Minoru Ishikawa and Yuichi Hashimoto *J. Med. Chem.* **54**, 1539-1554 (2011)

Solubility

MMP-12 inhibitors



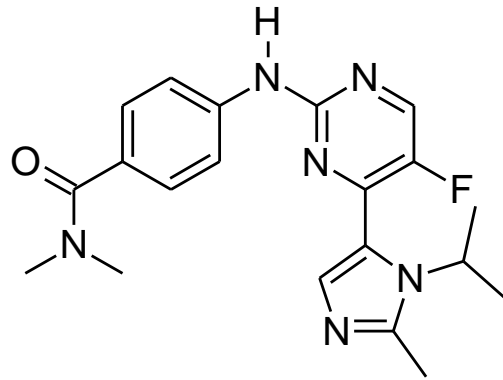
MMP-12 IC_{50} : 70 nM
s: < 1 $\mu\text{g/ml}$
clogP: 5.7



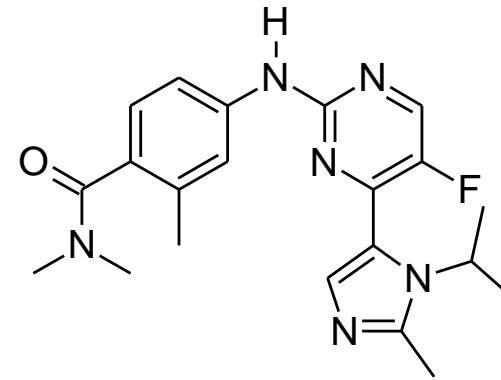
MMP-12 IC_{50} : 90 nM
s: 134 $\mu\text{g/ml}$
clogP: 4.6

Solubility

CDK2 inhibitors



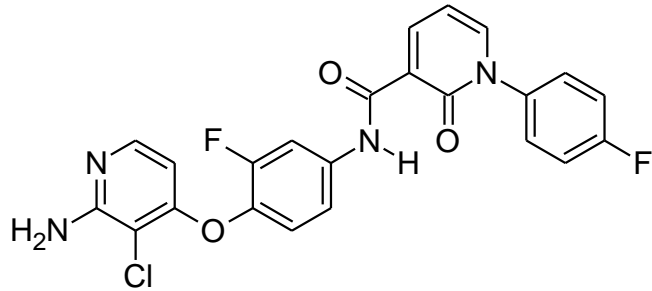
CDK-2 IC₅₀: 2 nM
s: 11 μM
clogP: 2.5



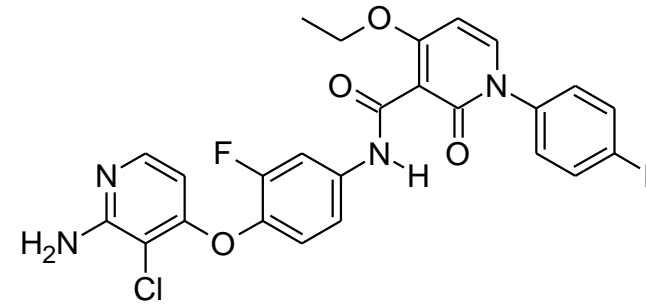
CDK-2 IC₅₀: 9 nM
s: > 2600 μM
clogP: 3.0

Solubility

Met kinase inhibitors



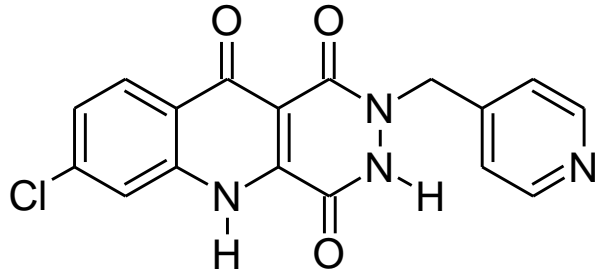
Met IC_{50} : 1.0 nM
s at pH 1: < 10 μ g/ml
clogP: 4.0



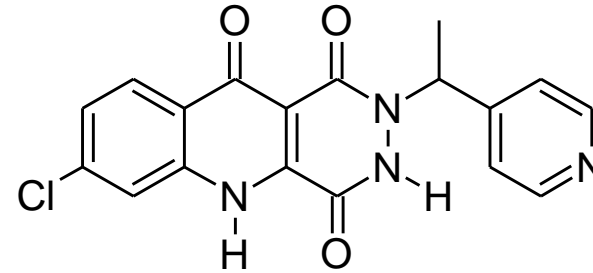
Met IC_{50} : 3.9 nM
s at pH 1: 400 μ g/ml
clogP: 4.4

Solubility

NMDA antagonists



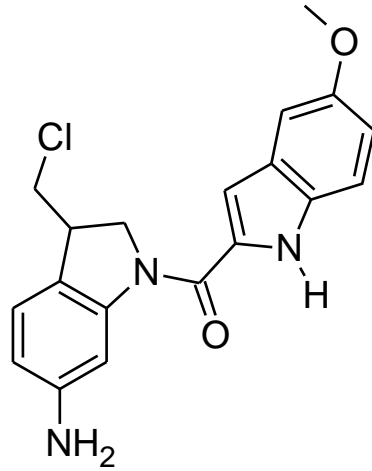
[³H]MDL 105519 K_i : 115 nM
s at pH 7.4: 0.05 mg/ml
Rat F: 5%
m.p.: 277-278 °C
clogP: 0.98



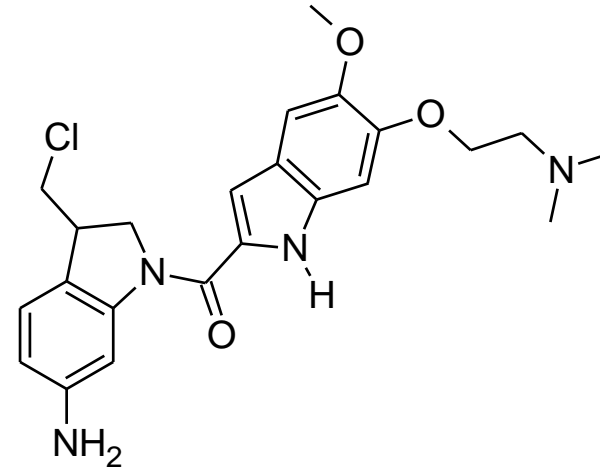
[³H]MDL 105519 K_i : 248 nM
s at pH 7.4: > 0.29 mg/ml
Rat F: 30%
m.p.: 245-247 °C
clogP: 1.3

Solubility

Antitumor agents



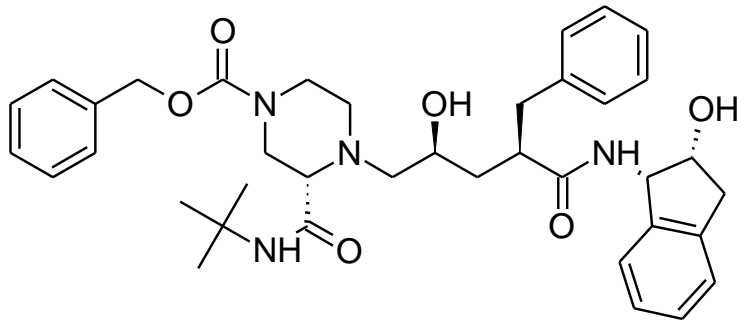
AA8 IC₅₀: 0.31 μM
s: 23 μM



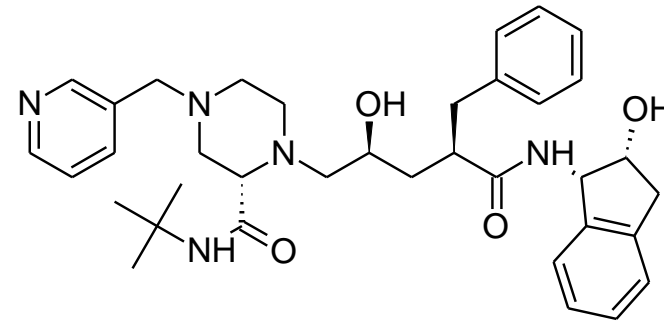
AA8 IC₅₀: 0.22 μM
s: > 1200 μM

Solubility

Protease inhibitors (indinavir)



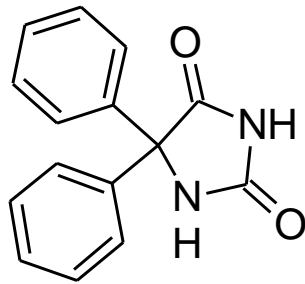
logP: 4.67
s: < 0.001 mg/ml
C_{max}: < 0.10 μM



logP: 2.92
s: < 0.07 mg/ml
C_{max}: 11.4 μM

Solubility

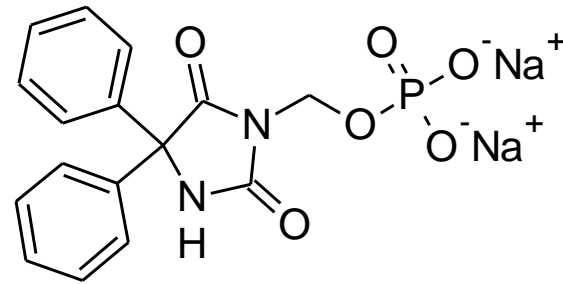
Prodrug for increasing solubility



Phenytoin

s: 20-25 $\mu\text{g/ml}$

Problematic formulation



Fosphenytoin

s: 142 mg/ml (4400 fold!)

Cerebryx™

**Metabolism,
metabolic stability,
prodrugs**

Metabolism describes the chemical reactions that change drugs into compounds which are easier to eliminate. The products of these chemical reactions are called metabolites.

Phase 1 metabolism involves chemical reactions such as oxidation (most common), reduction and hydrolysis. There are three possible results of phase 1 metabolism. The drug becomes completely inactive. In other words, the metabolites are pharmacologically inactive. One or more of the metabolites are pharmacologically active, but less so than the original drug. The original substance is not pharmacologically active, but one of its metabolites is. The original substance is called a prodrug.

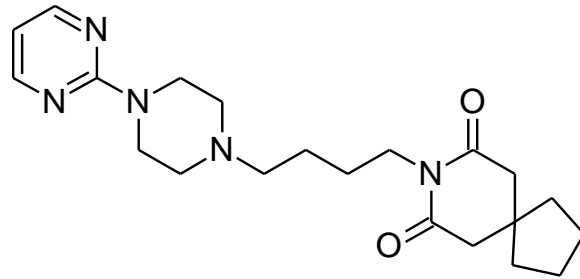
Phase 2 metabolism involves reactions that chemically change the drug or phase 1 metabolites into compounds that are soluble enough to be excreted in urine. In these reactions, the molecule (drug or metabolite) is attached to an ionisable grouping. This is called conjugation and the product is called a conjugate. Metabolites formed in phase 2 are unlikely to be pharmacologically active.

Structure Modification Strategies for Phase I Metabolic Stability Improvement

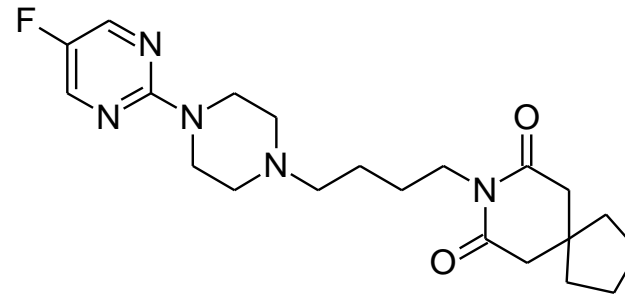
- Block metabolic site by adding fluorine
- Block metabolic site by adding other groups
- Remove labile functional group
- Cyclization
- Change the ring size
- Change chirality
- Reduce lipophilicity

Phase I Metabolic Stability

5HT_{1A} partial agonists (Buspiron/anxiolytic)

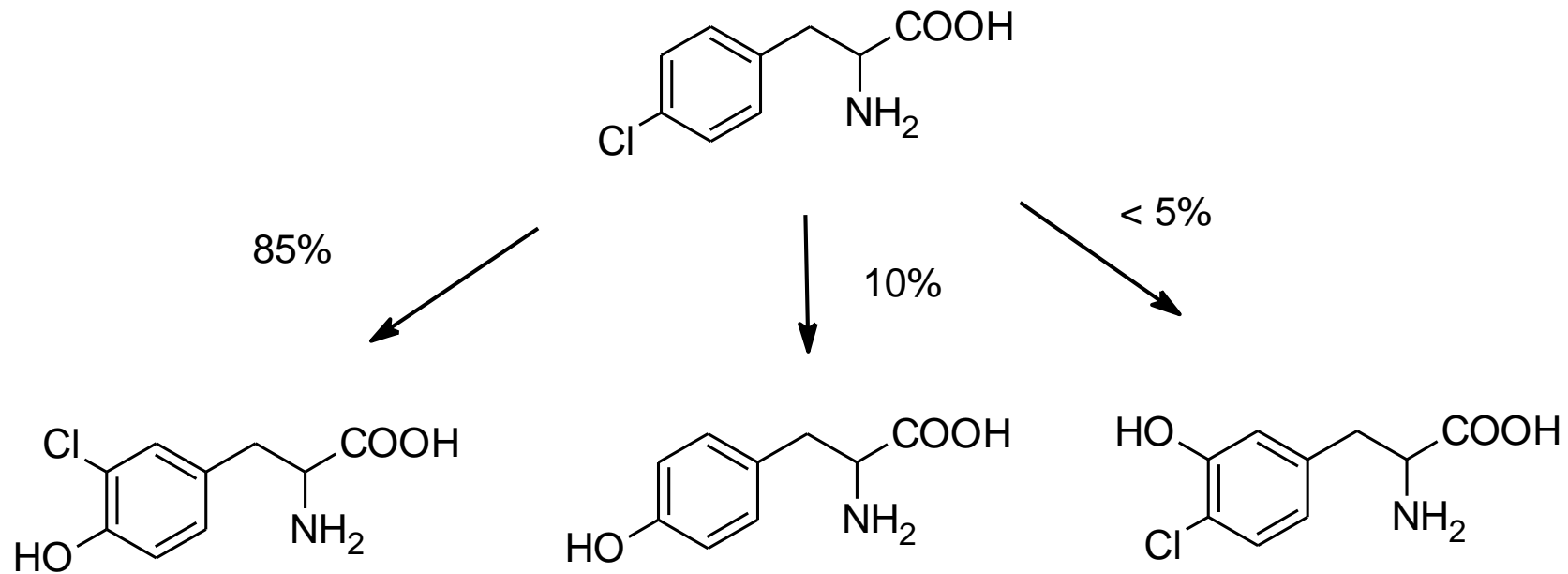


5HT_{1A} IC₅₀: 25 nM
CYP3A4 t_{1/2}: 4.6 min



5HT_{1A} IC₅₀: 63 nM
CYP3A4 t_{1/2}: 52.3 min

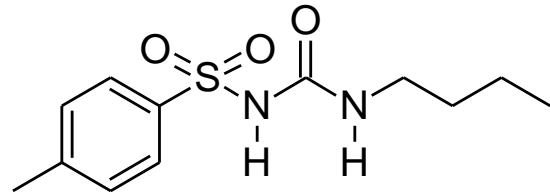
Phase I Metabolic Stability



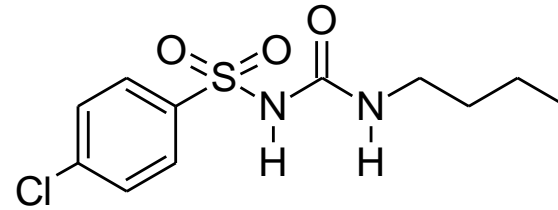
Gordon Guroff et al. *Science* **158**, 1524-1530 (1967)

Phase I Metabolic Stability

Hypoglycemics (Tolbutamide-Chlorpropamide)



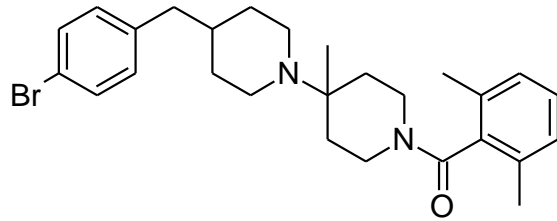
Clearance: 0.22 ml/min/kg
 $t_{1/2}$: 5.9 hr



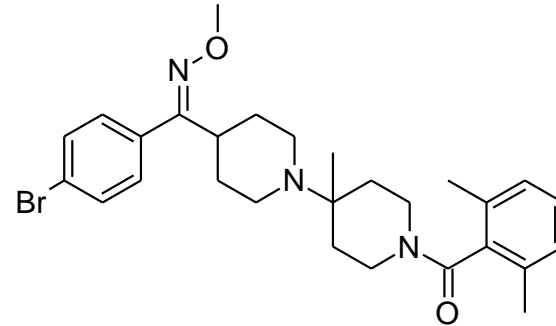
Clearance: 0.03 ml/min/kg
 $t_{1/2}$: 33 hr

Phase I Metabolic Stability

CCR5 receptor antagonists (anti-HIV activity)



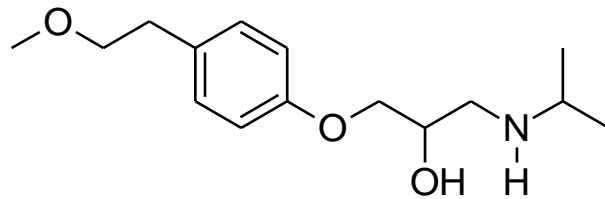
CCR5 IC_{50} : 10.0 nM
AUC (p.o.): 0.04 h.µg/ml



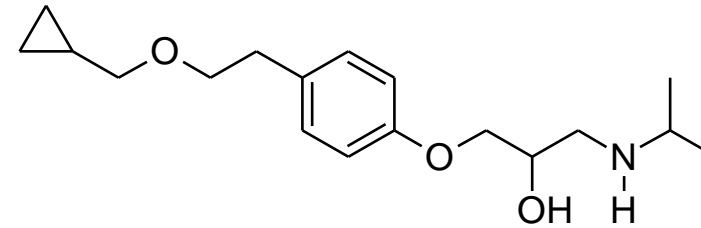
CCR5 IC_{50} : 1.3 nM
AUC (p.o.): 1.2 h.µg/ml

Phase I Metabolic Stability

β -blockers (metoprolol - betaxolol)



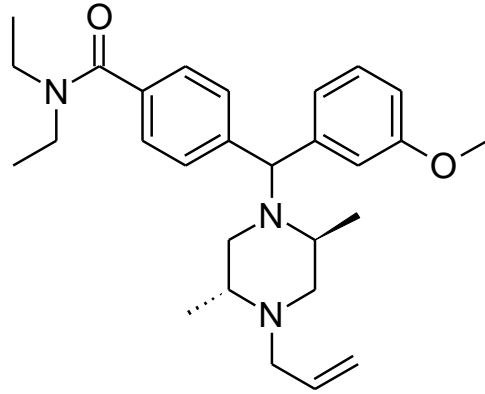
First Pass Elimination: 50%
Human $t_{1/2}$: 3.5 - 6 hr



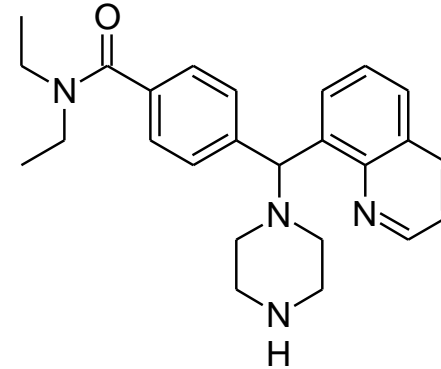
First Pass Elimination: 15%
Human $t_{1/2}$: 16 - 22 hr

Phase I Metabolic Stability

δ -opioid agonists



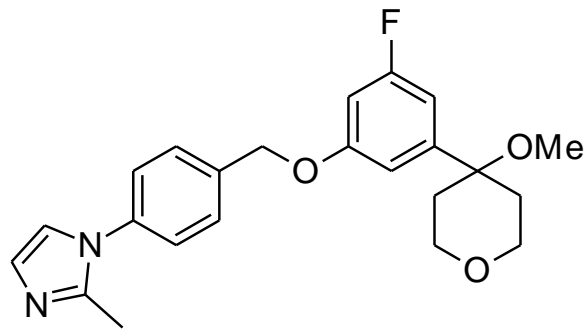
IC_{50} : 1.3 nM
Stability (rat): 1%



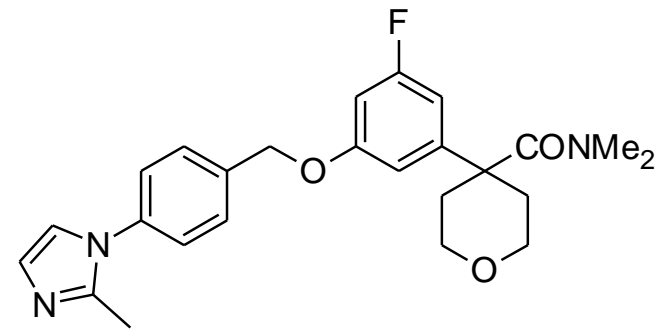
IC_{50} : 0.51 nM
Stability (rat): 52%

Phase I Metabolic Stability

5-lipoxygenase inhibitors



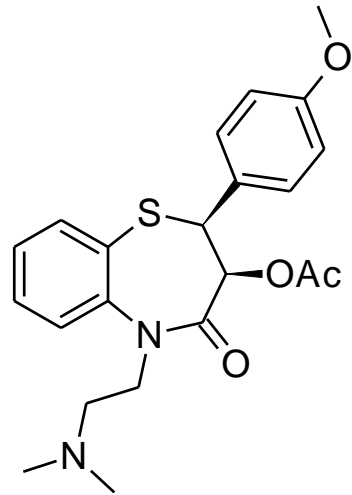
IC_{50} : 60 nM
 C_{max} : 0.24 $\mu\text{g/ml}$



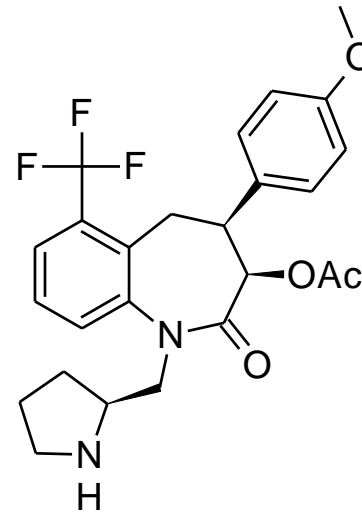
IC_{50} : 340 nM
 C_{max} : 1.57 $\mu\text{g/ml}$

Phase I Metabolic Stability

Calcium channel blockers



IC_{50} : 210 nM
Decrease in BP (0-6 h): 23%



IC_{50} : 91 nM
Decrease in BP (0-6 h): 45%

Structure Modification Strategies for Phase II Metabolic Stability Improvement

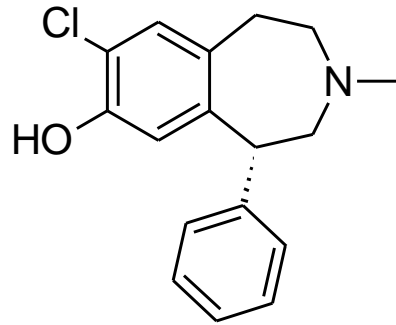
Introduce electron-withdrawing groups and/or steric hindrance

Change phenolic hydroxyl to cyclic urea or thiourea

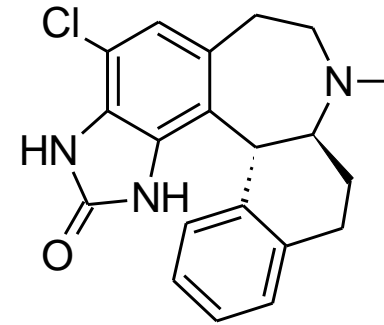
Change phenolic hydroxyl to prodrug

Phase II Metabolic Stability

Dopamine D₁/D₅ antagonists



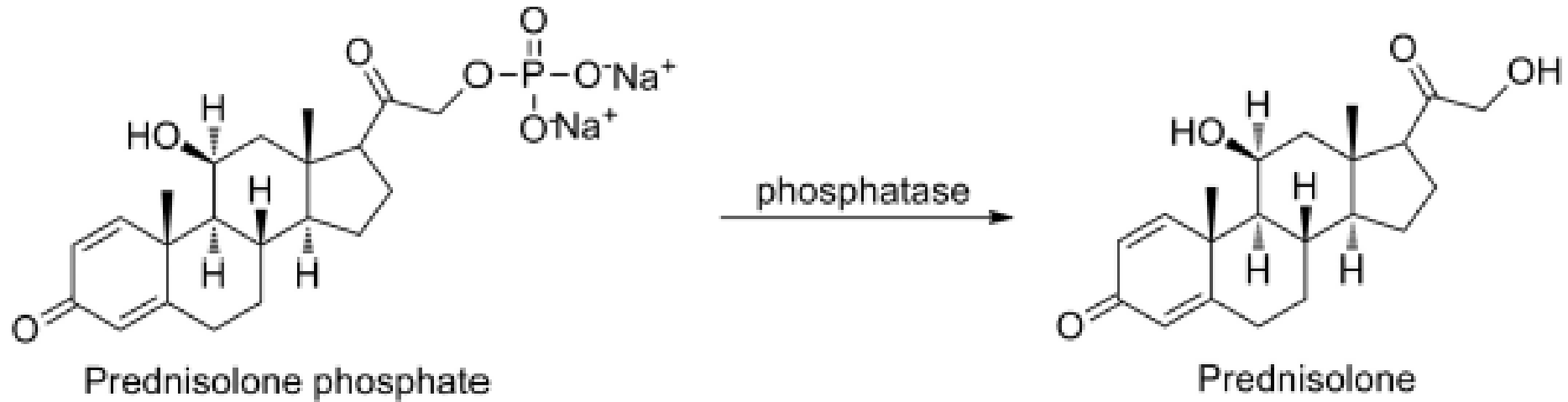
D₁ Ki: 1.2 nM
D₅ Ki: 2.0 nM
F: 0.6 %



D₁ Ki: 7 nM
D₅ Ki: 4.2 nM
F: 87 %

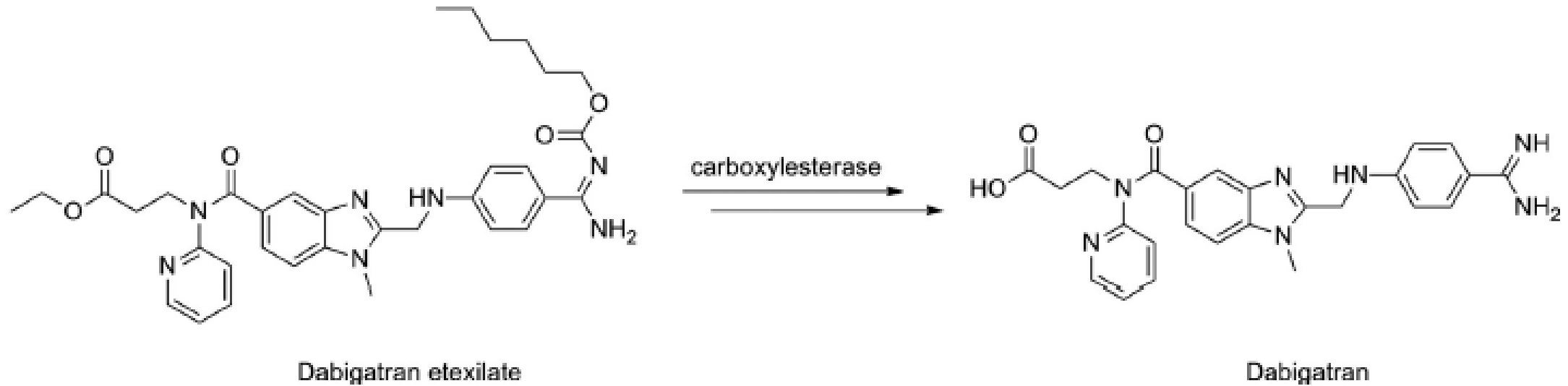
Prodrugs

Immunosuppressant, antiinflammatory ($s > 30$ times greater)



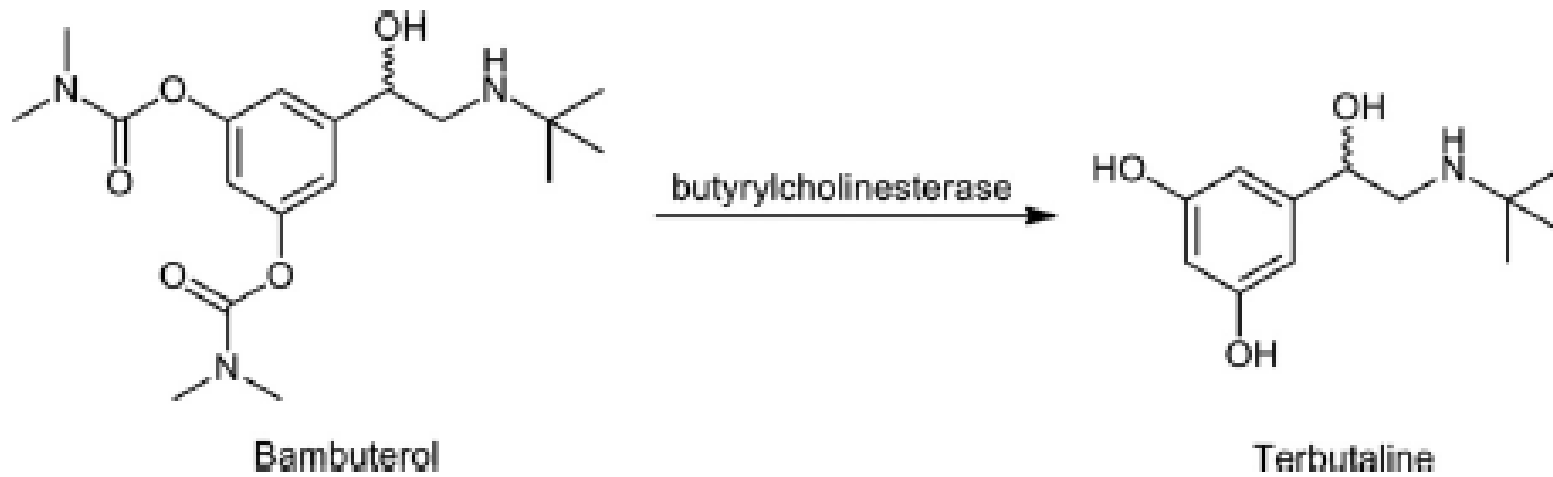
Prodrugs

Direct thrombin inhibitor for stroke prevention (BA: from 0% to 7%)



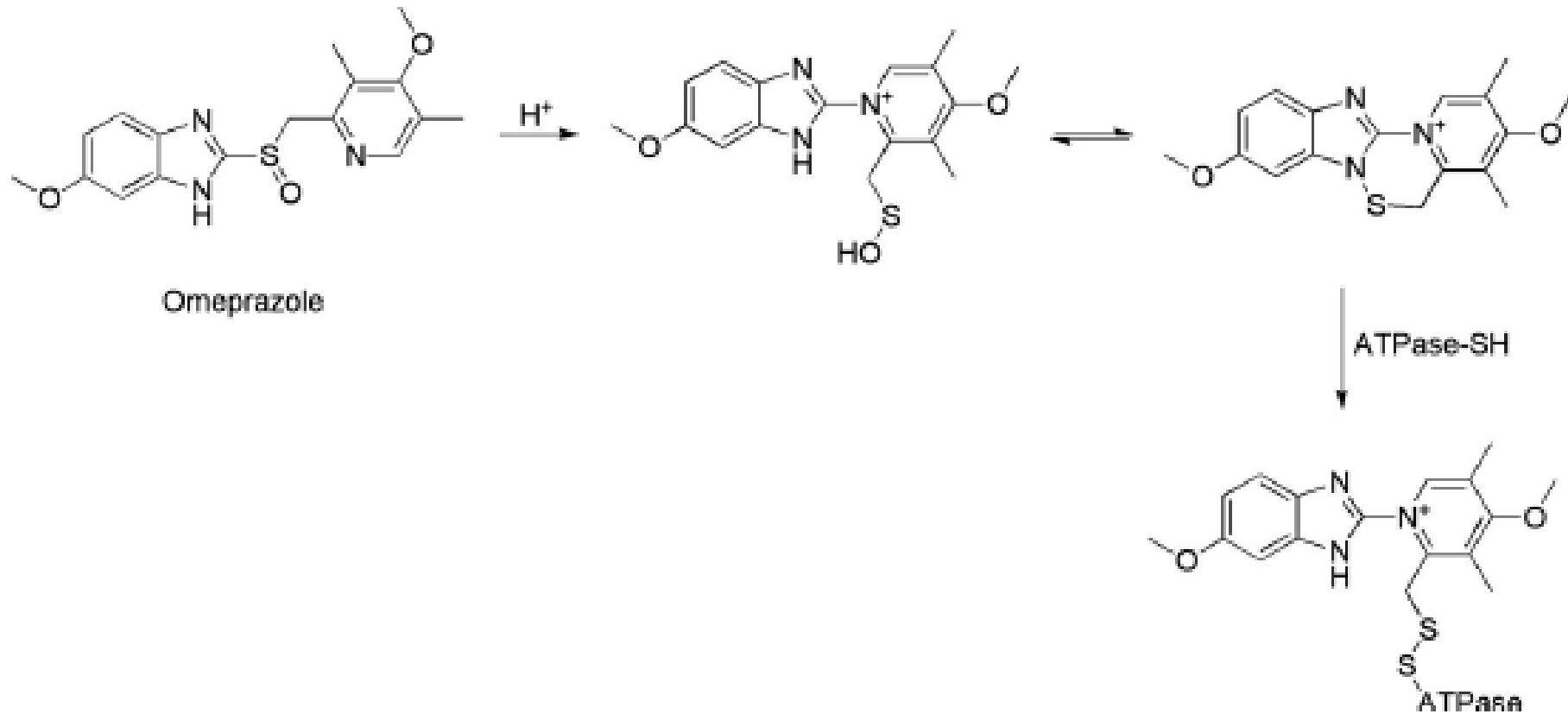
Prodrugs

β 2-agonist bronchodilator (once daily vs. three times daily)



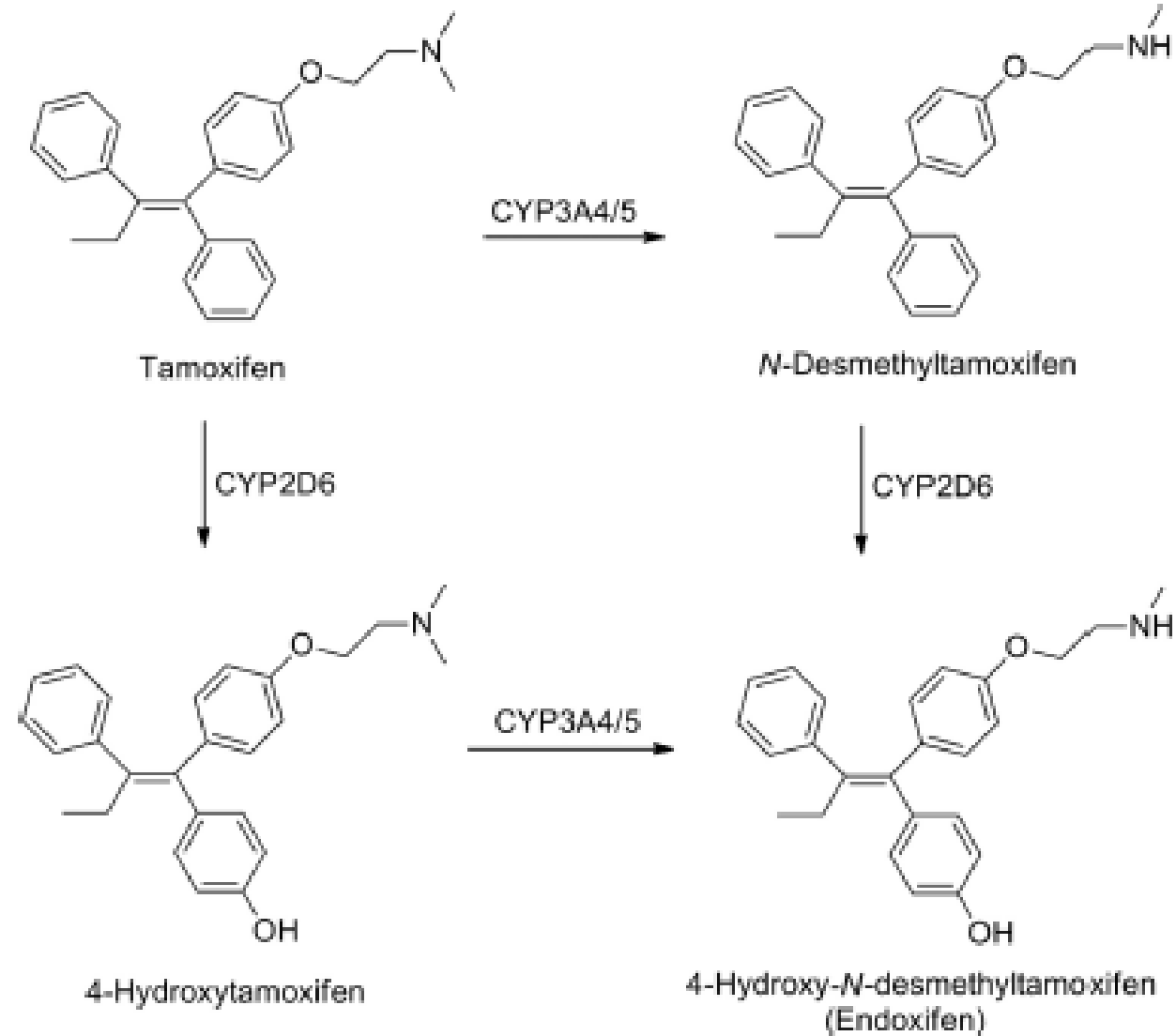
Prodrugs

Proton pump inhibitor (PPI) blocks H^+/K^+ ATPase



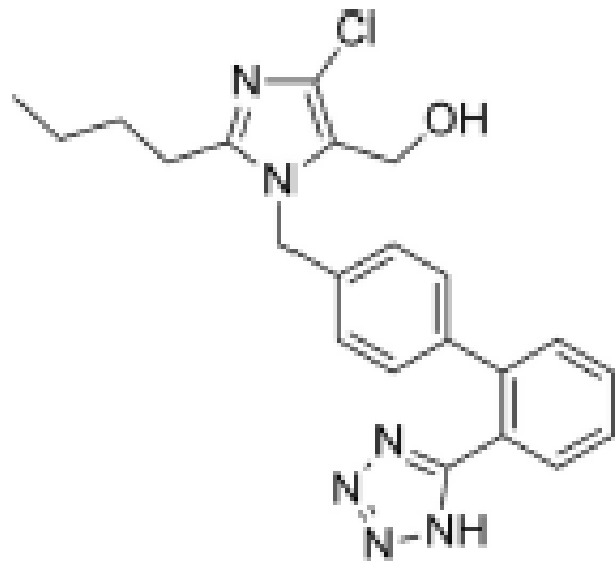
Prodrugs

Synthetic antiestrogen in breast cancer therapy

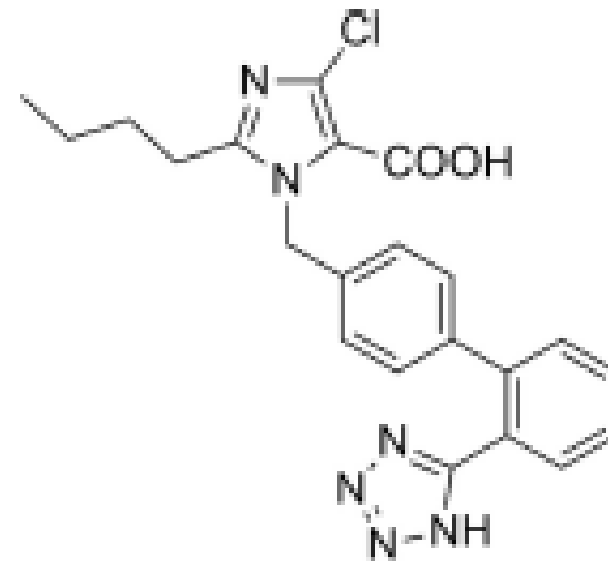


Prodrugs

Angiotensin II type 1 receptor antagonist (14% converted to EXP3174, 10-40 fold more potent)



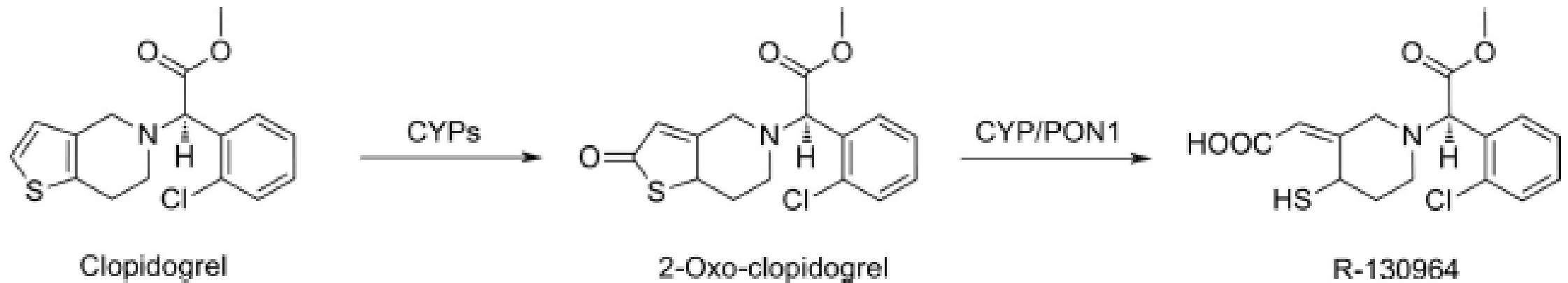
Losartan



EXP3174

Prodrugs

Antithrombotic (only 15% is transformed to the active R-130964 the rest turns to inactive acid)



**Scaffold hopping: practical application
of bioisosterism**

**Druglikeness, structure/property
relationships**

„Rule of 5“, CNS MPO

Aqueous solubility

Metabolism, metabolic stability, prodrugs

